



CHEM 8410_6410_4410 Spring 2019 – Mid-Term Exam 2
03-12-19

Time: 10:00am – 11:15am

Student Name: _____

Student Number: _____

Instructor:	Prof. Andreana
Room #:	BO 2059

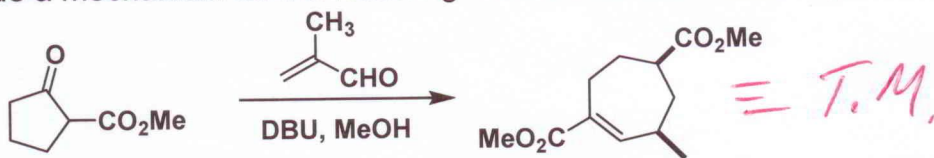
ANSWERS



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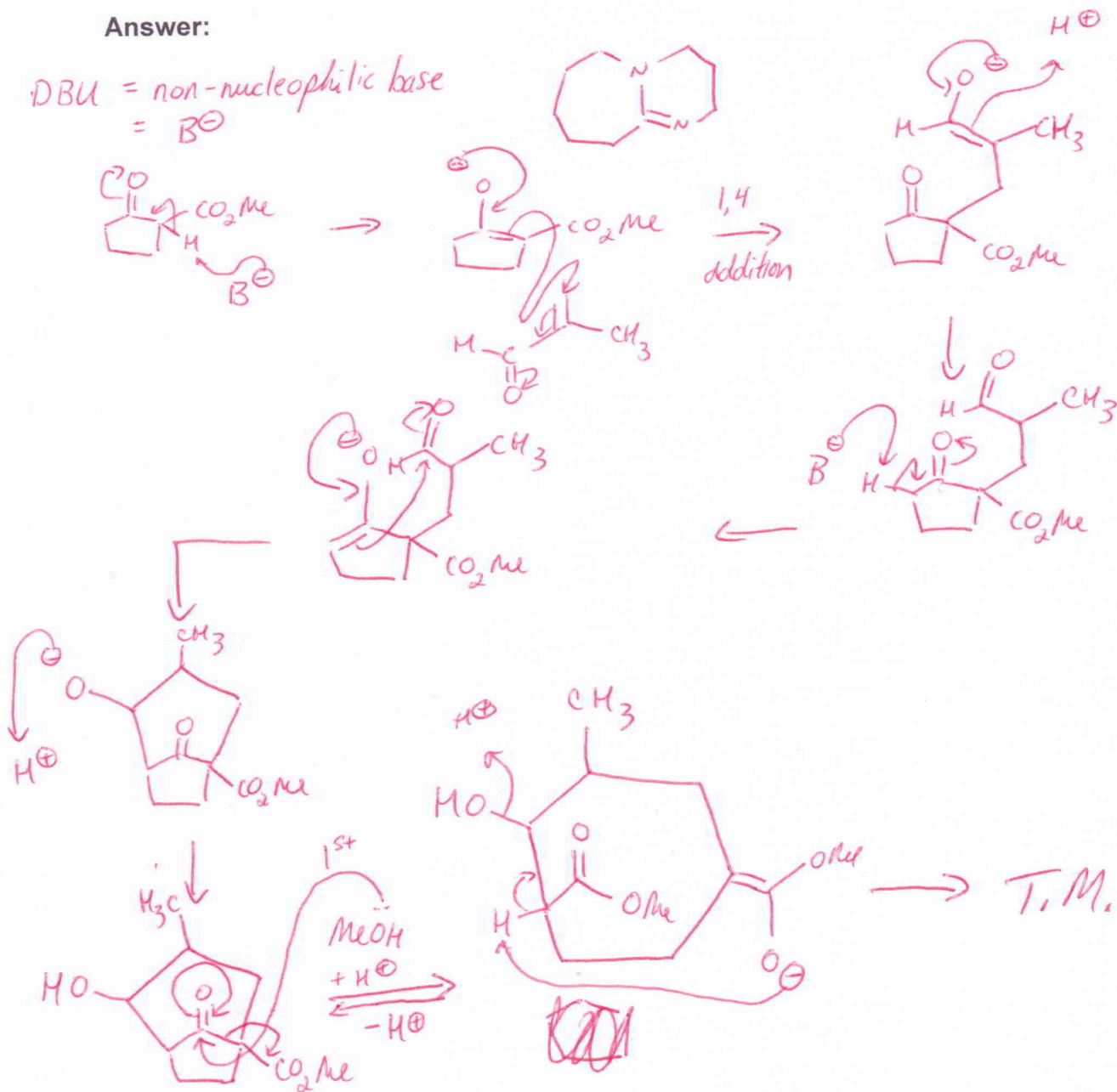


Problem 2: Please provide a mechanism for the following reaction transformation. (20 Points)



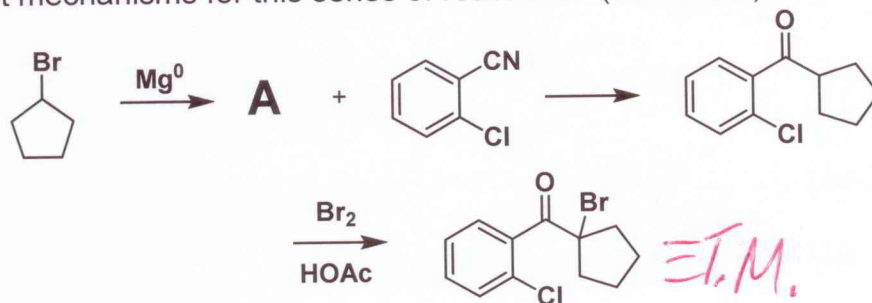
Answer:

DBU = non-nucleophilic base
= B^{\ominus}

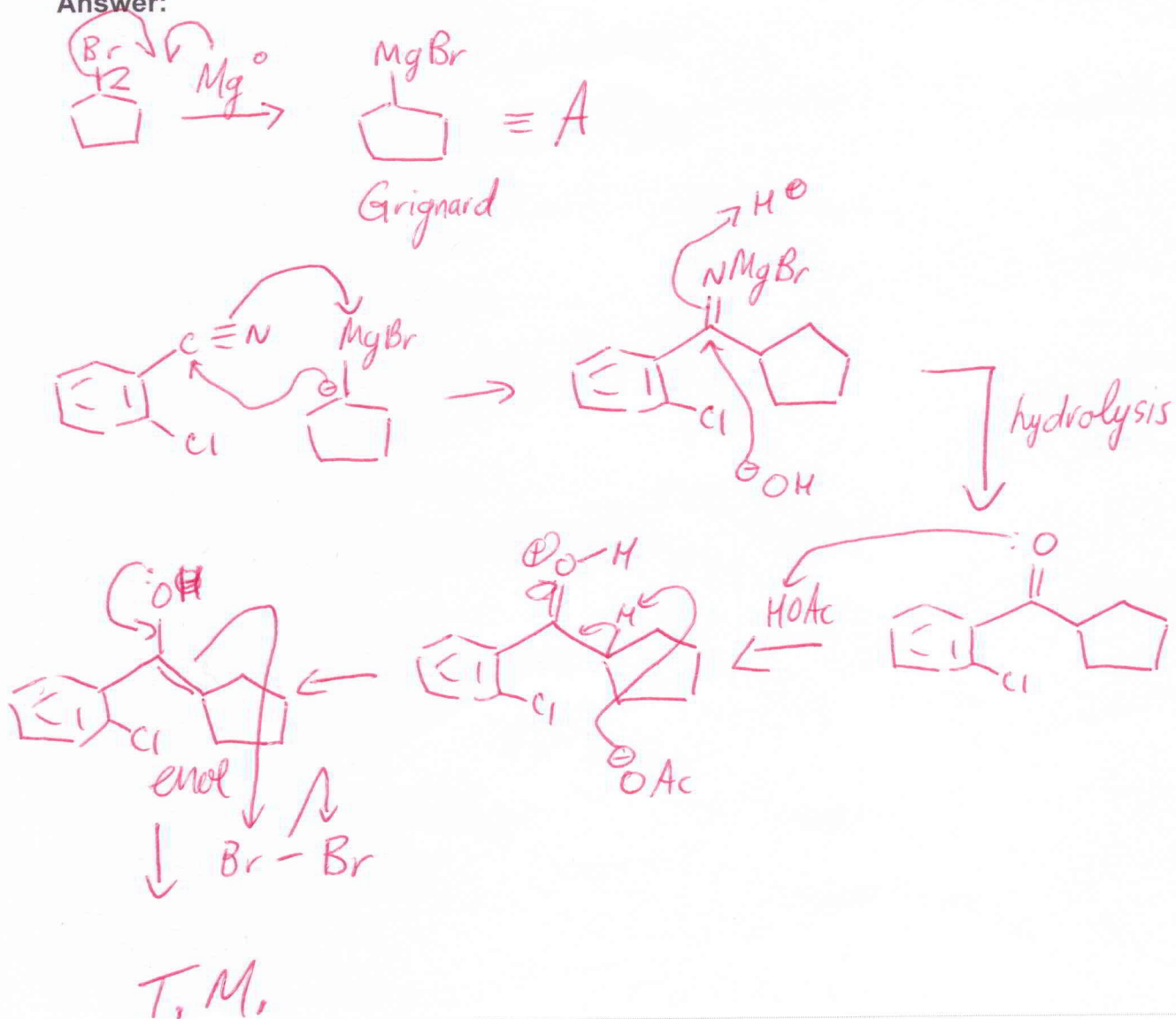




Problem 3: Suggest mechanisms for this series of reactions. (20 Points)



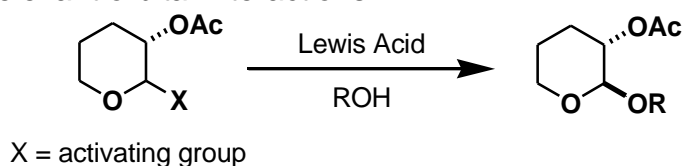
Answer:



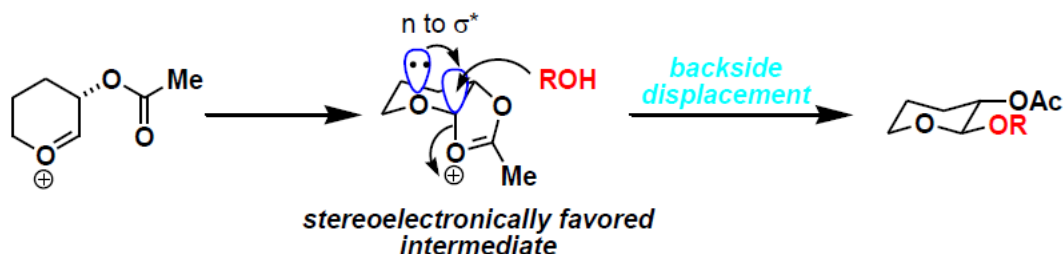


Problem 4: Stereoselective formation of the glycosidic linkage is the principal challenge in the synthesis of biologically important oligosaccharides. Anchimeric assistance (neighboring group participation) can be a powerful tool for the selective construction of glycosidic bonds. **(25 Points)**

Part A. For the following α -selective glycosylation, please provide a clear mechanism, using three-dimensional representations, that accounts for the observed stereochemical outcome. Indicate all relevant orbital interactions.

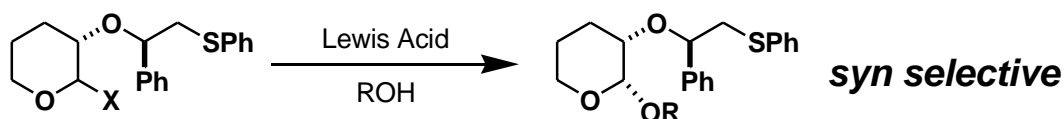


Answer:





Part B. Boons and co-workers (JACS, 2005) reported a highly selective synthesis of the corresponding syn di-substituted system by employing a participating phenyl-2-(phenylsulfanyl)ethyl moiety, as indicated below. Using three-dimensional drawings, provide a rational mechanism for this interesting reaction. Be sure to indicate all favorable and unfavorable interactions, both steric and electronic. N.B: This reaction is under kinetic control.



Answer:

