# PHYS 110 TECHNICAL PHYSICS STUDENT WORKBOOK

(3<sup>rd</sup> Edition revised Dec 2020) Professor Kevin Kimball, B.S.

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### ABOUT THIS WORKBOOK

This book is based on the formatted notebook model used by United States Navy class "A" technical schools. This format is a well proven and time-tested method of instruction; no fluff, no filler, and yet comprehensive and thorough.

For example, this method allows a Navy student to complete a *fully* transferrable undergraduate level three-credit course in Oceanography in two weeks!

# Key points in this format:

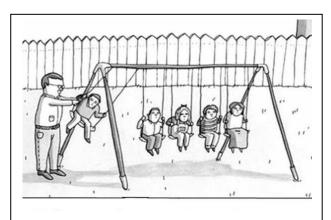
- 1. It minimizes the potential for ambiguity and enables the student to more effectively identify key points in a given lecture
- 2. It enables the student to more effectively "compare notes" with classmates
- 3. Both student and instructor are literally "on the same page."
- 4. Student ownership of specific course information is clearly delineated
- 5. It WORKS!

Exponent:

Scientific Notation: Provides a means			numbers.
Based on an understanding of	and	the	_
Algebra:		Scientific Notation:	
3 x <sup>2</sup>	3 x 1	L0 <sup>2</sup>	
The same rules pertaining to in Scientif			
Coefficient: Base:			
			Oog, the Cave Man

# Positive Exponents

Let's start with the easy stuff - multiplication:



Why physics teachers should not be given playground duty

Negative Exponents	
Bottom line:	
A negative exponent means you're dealing with a	or <b>a</b>

'Zero" power: $(x^0)$
Any non-zero number to the "zero power" equals
Rationale:
First power: $(x^1)$
Any number to the 1 <sup>st</sup> power equals
Standard Notation:

# Lab exercises:

3.8° =	
$3.8 \times 10^0 =$	
3.8 <sup>1</sup> =	
$3.8 \times 10^1 =$	
5 <sup>3</sup> = (standard notation)	
672 x 10 <sup>3</sup> = (standard notation)	
6 <sup>-2</sup> = (include both possible answers)	
In the expression "3.12 x 10 <sup>4</sup> " the "3.12 is called the	
9.36-1 =	
$x^5 \times x^7 =$	
$y^6 \div y^4 =$	
53 <sup>0</sup> =	
87 <sup>1</sup> =	
$2.38 \times 10^0 =$	
Definition of an exponent:	
In the expression "8.75 x 10 <sup>7</sup> " the "10" is called the	
A negative exponent indicates you're dealing with a or a	
$5^{15} \times 5^{-17} = $ (include all possible answers)	

 $\mathsf{L}$ 

Converting numbers in standard n	otation to scientific notation	
If the decimal point moves to the _	, the exponent go	es
the decimal point moves to the, the exponent goes		es
Converting numbers in Scientific N	lotation to Standard Notation	
f the exponent moves, the decimal point moves		
If the exponent moves	he exponent moves, the decimal point moves	
"Cheater Rule" for Standard Notat	ion:	
Any number in standard notation c	an be expressed as	times
General Rule for <u>correct</u> scientific	notation:	
"Only one in the	to the of the d	ecimal"
Very Important Exception:		
It is often more convenient to igno	ore this rule during	
(In other words, do	n't get hung up on this and create r	nore problems than necessary)
Example 1:	Example 2:	Example3:
Convert 873.463 into correct scientific notation	Convert 0.00785 into correct scientific notation	Convert 56.98 x 10 <sup>5</sup> into correct scientific notation

<u>Lab Exercises:</u> Page 7

# **Convert to correct scientific notation:**

186, 000	0.0045
5280	34.78 x 10 <sup>3</sup>
783.487 x 10 <sup>-8</sup>	2.85
0.000859 x 10 <sup>-9</sup>	0.0835 x 10 <sup>6</sup>
0.0000386 x 10 <sup>12</sup>	73.96
1/4	.937
Convert to standard notation:	

### **Convert to standard notation:**

1.63 x 10 <sup>3</sup>	3.637 x 10 <sup>-1</sup>
2.94 x 10 <sup>-6</sup>	36.345 x 10 <sup>2</sup>

# **Operations in Scientific Notation**

Multiplication Critical Rules:		
Multiply		
Retain		
Add		
NOTE:		
Adding a negative number is the same as _	 _ a	 
Example 1:		
$(4.75 \times 10^3) \times (2.43 \times 10^7)$		
Example 2:		
$(3.72 \times 10^7) \times 1.67 \times 10^{-2})$		

Division Critical Rules:	
Divide	
Retain	
Subtract	
NOTE: Subtracting a negative number is the same as a a	
Example 1:	
$(9.35 \times 10^8) \div (3.54 \times 10^4)$	

Example 2:

 $(8.62 \times 10^6) \div (3.97 \times 10^{-3})$ 

# **Addition/Subtraction Critical Rules:**

Exponents	_
Add/Subtract	
Retain	
Retain	

Example 1:

$$(6.72 \times 10^3) + (2.97 \times 10^3)$$

\_\_\_\_\_

Example 2:

$$(9.56 \times 10^5) - (8.47 \times 10^4)$$

# **Squaring and Cubing numbers in Scientific Notation - Critical Rules**

Square / Cube			
Retain			
Multiply	by	or	
RECALL:			
	s results in a		
Multiplying like signs re	esults in a		
Example 1:		Example 2:	
	2.56 x 10 <sup>3</sup> ) <sup>2</sup>	Example 2.	(2.56 x 10 <sup>3</sup> ) <sup>3</sup>
,	,		,
Example 3:	242 40-6.2	Example 4:	(2.4240-6.)3
(:	3.12 x 10 <sup>-6</sup> ) <sup>2</sup>		$(3.12 \times 10^{-6})^3$

<b>Square Roots/Cube Roo</b>	ts in Scientific Not	ation - Critical F	Rules
Exponent must be divisil	ole by	or	
Square/Cube root			
Retain			
Divide		or	
RECALL:			
Dividing unlike signs resu	ults in a		
Dividing like signs results			
Example 1: $\sqrt{9.46 \text{ x}}$	10 <sup>6</sup>	Example 2:	$\sqrt[3]{9.46 \times 10^6}$
Example 3:		Example 4:	
$\sqrt{8.1}$ x $^{\circ}$	10 <sup>5</sup>		$\sqrt[3]{8.1 \times 10^5}$

Lab Exercises Page 13

$(3.93 \times 10^7) \times (5.37 \times 10^6)$	Answer:
$(3.92 \times 10^3) \times (3.48 \times 10^{-5})$	Answer:
(3.92 × 10 / × (3.46 × 10 )	Allswei.
$(8.14 \times 10^7) \div (4.05 \times 10^9)$	Answer:
$(8.16 \times 10^{-5}) \div (4.89 \times 10^{6})$	Answer:
	•

$(3.93 \times 10^7) \times (5.37 \times 10^6)$	l
$3.93 \times 10^7$ 2.1104	4 x 10 <sup>14</sup>
	1 X 10
$\frac{5.37 \times 10^6}{1.013}$	
$21.104 \times 10^{13}$	
$= 2.1104 \times 10^{14}$	
(3.92 x 10 <sup>3</sup> ) x (3.48 x 10 <sup>-5</sup> )	
	x 10 <sup>-1</sup>
$3.48 \times 10^{-5}$	
$\frac{3.48 \times 10^{-5}}{13.642 \times 10^{-2}}$	
= 1.364 x 10 <sup>-1</sup>	
$(8.14 \times 10^7) \div (4.05 \times 10^9)$	
$ \begin{array}{c c} 8.14 \times 10^{7} \\ \hline 4.05 \times 10^{9} \end{array} $	8 x 10 <sup>-2</sup>
$= 2.0098 \times 10^{-2}$	
$(8.16 \times 10^{-5}) \div (4.89 \times 10^{6})$	
$\frac{8.16 \times 10^{-5}}{4.89 \times 10^{6}}$ 1.669	× 10 <sup>-11</sup>
= 1.669 x 10 <sup>-11</sup>	

$\sqrt{2.36 \times 10^7}$	
(0 (4 407) (4 07 407)	
$(2.64 \times 10^7) (1.37 \times 10^7)$	
$(3.93 \times 10^7)^2$	
(3.93 x 10 <sup>-</sup> )-	
$(5.37 \times 10^6)^3$	
$\sqrt{8.26 \times 10^8}$	
V0.20 X 10	
$\sqrt[3]{5.06 \times 10^{16}}$	
<b>V</b> 5.55 · · · 25	

	1
$\sqrt{2.36 \times 10^7}$ Change to : $\sqrt{23.6 \times 10^6}$ (exponent divisible by 2)	4.858 x 10³
$= 4.858 \times 10^3$	
$(2.64 \times 10^7) - (1.37 \times 10^7)$	
$2.64 \times 10^{7}$	$1.27 \times 10^7$
$-1.37 \times 10^{7}$	
$1.27 \times 10^{7}$	
$(3.93 \times 10^7)^2$	
$= 15.445 \times 10^{14}$	1.5445 x 10 <sup>15</sup>
$= 1.5445 \times 10^{15}$	
$(5.37 \times 10^6)^3$	
$= 154.854 \times 10^{18}$	1.5485 x 10 <sup>20</sup>
$= 1.54854 \times 10^{20}$	
$\sqrt{8.26 \times 10^8}$	
2.874 x 10 <sup>4</sup>	2.874 x 10 <sup>4</sup>
$\sqrt[3]{5.06 \times 10^{16}}$	
γ 3.00 Λ 10	2.600 105
change to: $\sqrt[3]{50.6 imes10^{15}}$ (exponent divisible by 3)	3.699 x 10⁵
$= 3.699 \times 10^{5}$	

# **Metric System**

# Major advantage of the metric system:

It can be applied directly to			
Uses a system of and			
Units relate to specific	("whatcl	na got")	
Prefixes are specific	("ho	w many you got")	
Prefixes are mathematically	w	ith o	of
Examples of units:	Example	es of prefixes:	
UNIT: MEASUREMENT OF:	PREFIX:	EQUIVALENT:	
Meter	Centi	or	
Liter	Milli	or	
Gram	Kilo	or	
Hence:			
8.5 centimeters = 8.5 x m	neters		
500 milliliters = 500 x lite	ers		
6.75 kilograms = 6.75 x §	grams		
BOTTOM LINE:			
Any prefix can be replaced or subs	tituted with a	of	

# COMMONLY USED METRIC PREFIXES: (Required knowledge!)

Pre	efix:	/ Standard notation:	/ Fraction:	/ Power of ten:
	giga			
	mega			
	kilo			
	centi			
	milli			
	micro			

# self check:

centi = (standard notation)	
kilo = (power of ten)	
10 <sup>9</sup> = (prefix)	
10 <sup>-3</sup> = (standard notation)	
0.01 = (prefix)	
giga = (power of ten)	
1000 = (power of ten)	
10 <sup>6</sup> = standard notation)	
milli = (power of ten)	
micro = (standard notation)	
10 <sup>-2</sup> = (prefix)	
0.001 = (power of ten)	
10 <sup>3</sup> = (prefix)	
mega = (standard notation)	
milli = (standard notation)	
1,000,000 = (power of ten)	
0.000001 = (power of ten)	
kilo = standard notation	
10 <sup>6</sup> = (prefix)	
1,000,000,000 = (power of ten)	

**Standard Notation** 

# **Informal Lab Exercise:**

8.97 kilograms = ? grams		
6.5 centimeters = ? meters		
7.5 gigavolts = ? volts		
4.7 microFarads = ? Farads		
6.4 megawatts = ? watts		
9.87 mililiters = ? liters		
Quantity:	 Equivalent q	uantity with prefix
5483 grams		
0.0268 meters		
9,700,000,000 volts		
0.0000056 Farads		
0.0045 liters		
4,300,000 watts		

Scientific Notation

Conversion hack:

Convert 3.567 x 10<sup>5</sup> gigavolts to millivolts

5.98 x 10 <sup>4</sup> kilograms = _?_ grams	
8.34 x 10 <sup>-1</sup> meters = _?_ centimeters	
5.92 x 10 <sup>-4</sup> megawatts = _?_ watts	
(Answer in <u>standard</u> notation:)	
500 millivolts = _?_ volts	
345 grams = _?_ kilograms	
6.73 x 10 <sup>5</sup> centimeters = _?_ meters	
$3.81 \times 10^{-4} \text{ volts} = _?\_ \text{ microvolts}$	
(Answer in correct scientific notation:)	
3.45 x 10 <sup>5</sup> microvolts = _?_ kilovolts	
7.93 x 10 <sup>-5</sup> kilograms = _?_ milligrams	
5.78 x 10 <sup>3</sup> millimeters = _?_ centimeters	
4.32 x 10 <sup>3</sup> gigahertz = _?_ megahertz	

Displacement:	
Definitio	n(s):
	1and
	2 and
Symbol: _	
Standard	units:
	1. British ("U.S Standard"):
	2. Metric:
NOTE: In this co	ontext "displacement" does NOT refer to
Force:	
Definitio	
	1 or a
	2. That which may
Symbol: _	(Weight is a measure of)
Standard	units:
	1. British ("English"):
	2. Metric:
NOTE:	

Mass:				
Definition(s	s):			
	1			
	2	*		
	3			
* "	" : resista	ance to a	in	
Standard u	nits:			
	1. British:	( <u>not</u>	)	
	2. Metric:	/ (not	)	
Volume:				
Definition:	4			
	1			
Standard u	nits:			
	1. British:	(	)	
	2. Metric:	(	)*	
* NOTE: "	are also frequ	ently used to m	easure volume	e in metric terms
	considered as "standa			
J				
Time:				
 Definition:				
1. "That w	hich we			"
	,			
	-			
Standard ι	unit: (both British and r	metric)		
	1 ( N	IOT or		_)

# self check:

# Answers:

a measure of inertia	
that which we measure with a clock	
distance and direction	
standard metric unit of force	
standard British unit of displacement	
a quantity of space	
standard metric unit of mass	
a push or a pull	
length and direction	
standard British unit of force	
standard metric unit of volume	
that which may affect motion	
weight is a measure of _?_	
resistance to a change in motion	
standard British unit of volume	
standard British unit of mass	
standard unit of time	
standard metric unit of displacement	
a quantity of material	
stuff	

# **CONVERSIONS**

Based on the principles used in	, and
exploit the rules used in "	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,
Example 1:	
3 1 _	
$\frac{1}{4} \times \frac{1}{3} =$	
Example 2:	
$\frac{a}{x} \times \frac{b}{x} =$	
$\frac{-}{c} \times \frac{-}{a} =$	
Example 3:	
$\circ$ $\triangle$	
$\frac{1}{\Box} \times \frac{1}{\Box} =$	
Rationale:	

# Applying method of multiplying fractions as a means of converting units (The "factor-labeling" method)

Exam	рĺ	e	1	:
LAGIII	ρ.	_	_	

To convert <u>35 miles per hour</u> to <u>"x" feet per second</u>:

Step 1: Restate 35 MPH in fraction form:

1 hour

Step 2: Set up multiplication problem in fraction form so that the terms you

wish to change will be \_\_\_\_\_ -\_\_\_:

$$\frac{35\,mi}{1\,hour}\,X\,\frac{hour}{mi}\,X\,\frac{hour}{}$$

Step 3: Replace terms with those you want:

$$\frac{35 \, mi}{1 \, hour} \times \frac{feet}{mi} \times \frac{hour}{seconds}$$

Step 4: Inert correct mathematical equivalences:

$$\frac{35 \, mi}{1 \, hour} \, \mathsf{X} \, \frac{5.28 \, \mathsf{X} \, 10^3 \, feet}{1 \, mi} \, \mathsf{X} \, \frac{1 \, hour}{3.6 \, \mathsf{X} \, 10^3 sec}$$

-----

Step 5: Cross-cancel terms:

$$\frac{35 \, \textit{miles}}{1 \, \textit{hour}} \, \mathsf{X} \, \frac{5.28 \, \mathsf{x} \, 10^{3} \, \textit{feet}}{1 \, \textit{mile}} \, \mathsf{X} \, \frac{1 \, \textit{hour}}{3.6 \, \mathsf{x} \, 10^{3} \, \textit{sec}}$$

Step 6: Restate with remaining terms:

Step 7: Perform normal calculations one operation at a time until you reach an answer in the desired terms \*

$$\frac{35 \times 5.28 \, feet}{3.6 \, sec} = \frac{184.8 \, feet}{3.6 \, secs} =$$

\_\_\_\_\_ \* <u>ft/sec</u> (final answer)

Conversion factors you should know:			<del>-</del>	
	1 mile = 5.28 x 10 <sup>3</sup> ft	1 mile = 1.61 x 10 <sup>3</sup> meters	1 kilometer = 10 <sup>3</sup> meters	1 hour = 3.6 x 10 <sup>3</sup> seconds

Self -check: conversions Answers: Convert 400 MPH to ft/sec Convert 80 meters/sec to miles/hour Convert 220 ft/sec to MPH Convert 88 kilometers/hr to meters/sec

### Example 1

Given: "12", "3", and "4"

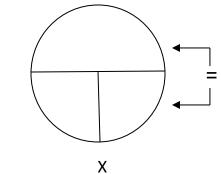
then:

$$\frac{12}{3} = 4$$

$$\frac{12}{4} = 3$$

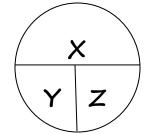
and 
$$3 \times 4 = 12$$

Or:



Example 2

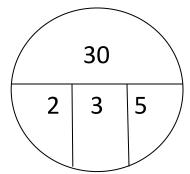
Given:



Then:

Example 3

Given:



Then: 30 = \_\_\_\_\_

"People, listen up!
There is the Right Way,
there is the Wrong Way,
and then there is the Navy Way,
and you <u>better</u> start learning the Navy Way!"

-Boatswain's Mate Second Class Donald Barger, USN, Navy Boot Camp Company Commander



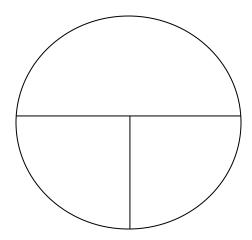
Example 4:

Given: 
$$\frac{ab}{cde} = fg$$
 Solve for "d"

Traditional solution:

Using the "egg"

$$\frac{ab}{cde} = fg$$
, Solve for "d"



*d* =

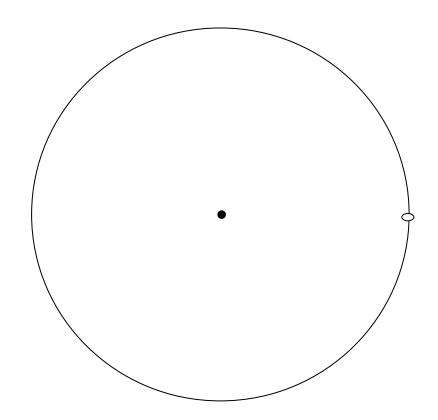
Self check:

ation:	"Egg"	Solution ( "n" = ?)
F = ma		m =
$P = \frac{W}{T}$		T =
$2as = (V_f^2 - V_i^2)$		a =
$KE = \frac{1}{2}mv^{2}$ (Hint: $\frac{1}{2} = .5$ )		m =
$D = R \times T$		T =
PE = mgh		h =
$A = (\cos \theta)(H)$		H =
$a = \frac{V_f - V_i}{t}$		t =

Structure of atom - a key to understanding "mass"

Example 1:

Hydrogen:



D	r	٦t	$\sim$	n	•

1 . \_\_\_\_\_ charge

2. Has approximately \_\_\_\_\_\_ times the mass of electron

### Electron:

1. \_\_\_\_\_charge

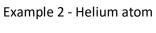
### Hydrogen:

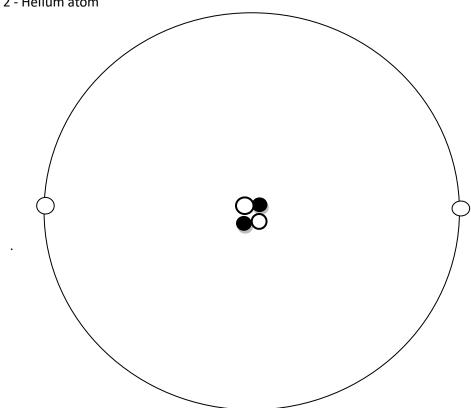
1. \_\_\_\_\_ of all atoms

2. \_\_\_\_\_\_ element in the universe

"

- Frank Zappa





(NOT to scale!)

1	N			_	ما		_	
	N	ш	11		$\boldsymbol{\rho}$	ľ	C	٦

1.	Contains		and	
----	----------	--	-----	--

2. Accounts for	of atom's	mass
	 -	

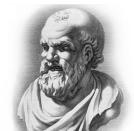
# Neutron:

2. \_\_\_\_\_ charge

1. Slightly more	than a proton, hence it also has
approximately _	times the mass of electron

By vo	olume, an atom is over	percent
Dy vc	name, an atom is over	percent

# **Atom Model History**



Democritus - Fifth century B.C.	
1 . All matter is composed of	
2. "Atom" : Greek for ""	
John Dalton - 1803	
1. Atom is a	
(AKA the " model")	
2. Each element was composed of	9
3. Different elements composed of	
4. Compounds are composed of atoms in	
5. Chemical reactions are of	,
And mass is therefore	
Joseph John Thompson - 1897	
1. "Plum":	
a. A sphere of diffuse electricit	y with
negative imbedded throughout	

2. Discovered \_\_\_\_\_\_\_, and was awarded \_\_\_\_\_\_ in 1906

# The "Solar System" Model



# **Ernest Rutherford - 1911**

1. Discovered that the atom is mostly	with a
	With a
dense charged surroun	ided by
negative	
Neils Bohr - 1913	0
1. Electrons travel in	
2. Only allowed	
3. Modern of the	Everything we call real is made of things
Electron Cloud Model - 1920's	cannot be regarded as real."
1. Erwin Schrodinger <sup>1</sup> and Werner Heisenburg <sup>2</sup>	~Niels Bohr
Developed functions to determine regions	s or clouds in
which are most likely to be found	
2. Heisenberg: Developed the Principle : I	
predict of single electron	

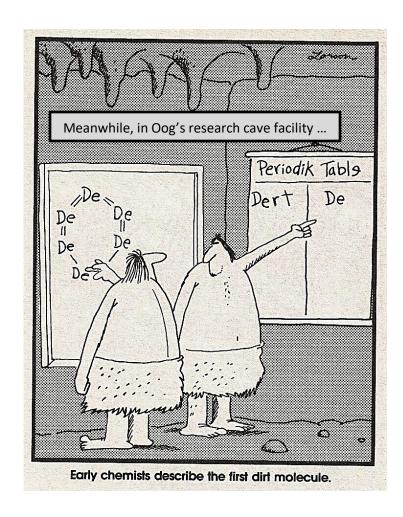
James Chadwick - 1932



1. British experimental physicist credited with discovering the \_\_\_\_\_

### Particles and average radii:

Particle	Approx. Radius
	10 <sup>-9</sup> meters
	10 <sup>-10</sup> meters
	10 <sup>-15</sup> - 10 <sup>-14</sup> meters
	10 <sup>-15</sup> meters
	10 <sup>-18</sup> meters



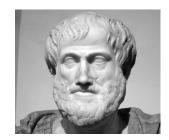
# More History: How We Got Here

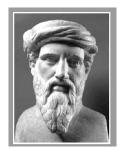
The 3-legged stool of understanding is held up by history, languages, and mathematics. Equipped with these three you can learn anything you want to learn. But if you lack any one of them you are just another ignorant peasant with dung on your boots.

- Robert A. Heinlein, author, engineer, U.S. Naval Academy graduate, curmudgeon.

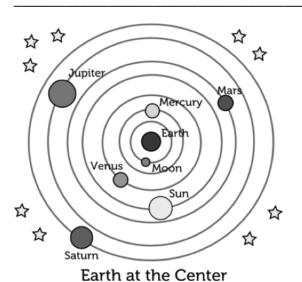
### Aristotle

1	 	 	
2			
3.			
4.			





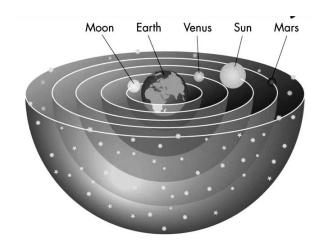
### **Pythagoras**

Ptolemy	Sphere of Stars Saturn  Jupiter  Wenus  Mercury  Moon
Epicycles	

Deferents

**Crystal Spheres** 



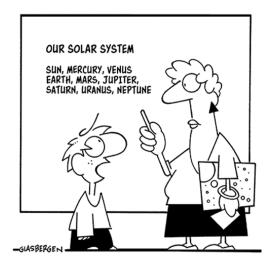
### Copernicus

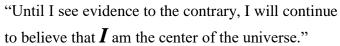


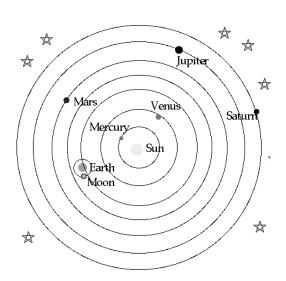
 1.

 2.

 3.

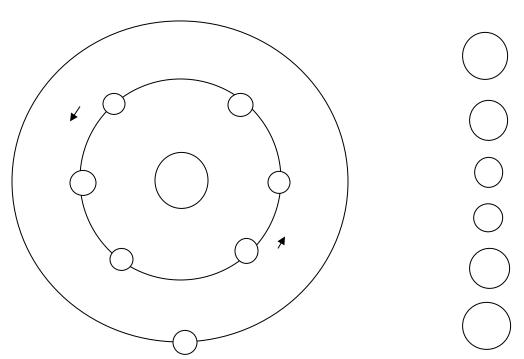


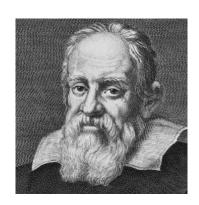




### Galileo

1. First to use the	
2. Discovered the	_ of
3. Discovered the	_ of
4. Discovered	
4. Prime author of the	
5. Was tried for	



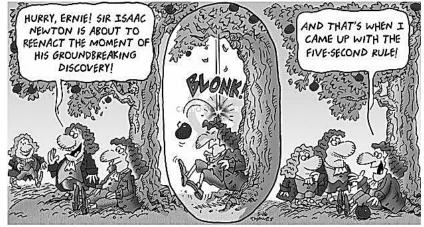


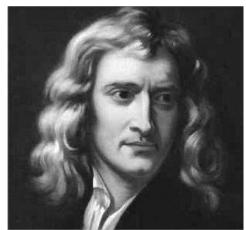


GALILEO DESCRIBES HIS DISCOVERIES TO THE CHURCH

#### **Sir Isaac Newton:**

1. Wrote	
2. Discovered	
3. Established the link	
4. Sought to	
5. Emphasized A,,	
6. Invented	
7. Derived planetary motions	





#### The Scientific Method

"Physicists are conservative revolutionaries. They do not give up tried and tested principles until experimental evidence - or an appeal to logical and conceptual simplicity - forces them into a new and sometimes revolutionary viewpoint. Such conservatism is at the core of the critical structure of inquiry. Pseudoscientists lack that commitment to existing principles, preferring instead to introduce all sorts of ideas from the outside."

- Dr. Heinz R. Pagels, "The Cosmic Code"

If it walks like duck, swims like duck, quacks like duck . . . it's probably not an elephant."

- Chief Petty Officer Ralph Caraway, Master Instructor, USN-retired, explaining the overarching theory of acoustic intelligence analysis.

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You can observe a lot just by watching."

-Yogi Berra, American philosopher

The Scientific Method is a remarkably adaptable tool that allows us "mere mortals" to pursue the most profound truths. Its strength lies in both its beautifully articulated process and its flexibility.

We keep the Scientific Method around because it works, and most importantly, it has **never failed**. Not even once. Its self-correcting nature prohibits failure.

Now that's a pretty bold if not outrageous statement, so let's bring the topic into sharper focus by stipulating a distinction between the "Scientific Method" and "Science" itself:

While the Scientific Method does not fail, Science often does. It happens all the time, and is a normal, entirely expected part of the business. The Scientific Method gives us the means to (1) recognize and deal with these failures and (2) establish the credibility of successes through a rigorous, clearly defined vetting process.

In short, the Scientific Method is how we police the business of Science.

Though frequently viewed as an esoteric, intellectual protocol, it also has very practical, down-to-earth applications. One beautiful example of this (I believe) is the grand experiment of American Democracy. People a lot smarter and more credentialed than me have long argued that it's no coincidence that the architects of the American government were also products of the Galilean/Newtonian revolution of scientific rationale (think Thomas Jefferson and Benjamin Franklin, both well-established scientists, inventors, and philosophers in their own right). Look closely, and you will see a remarkable similarity between the Scientific Method and our constitutional system of informed candid debate, peer review, accountability and a formal regimen of "checks and balances."

Both protocols are ultimately beholden to unvarnished reality, and survive the most rigorous challenges to their very existence because they are specifically engineered as fluid, adaptive processes of deliberative, critical analysis and self-correction.

"Galileo was one of the first people to practice what we recognize today as the scientific process (or "method"): the dynamic interplay between experience (in the form of experiments and observations)

and thought (in the form of creatively constructed theories and hypotheses). This notion that scientists

learn not from authority or from inherited beliefs but rather from experience and rational thought is what makes Galileo's work, and science itself, so powerful and enduring.

"Galileo's methods have been crucial to science ever since. They included:

- Experiments, designed to test specific hypotheses
- · Idealizations of real-world conditions, to eliminate (at least in ones's mind) any side
- effects that might obscure the main effects
- · Limiting the scope of inquiry by considering only one question at a time. For example,
- Galileo separated horizontal from vertical motion, studying only one of them at a time.
- · Quantitative methods. Galileo went to great lengths to measure the motion of bodies.
- · He understood that a theory capable of making quantitative predictions was more
- powerful than one that could make only descriptive predictions, because quantitative predictions were more specific and could be experimentally tested in greater detail

"Observation refers to the data gathering process. A measurement is a quantitative observation, and an **experiment** is an observation that is designed and controlled by humans, perhaps in a laboratory.

"A scientific **theory** is a well- confirmed framework of ideas that explain what we observe.

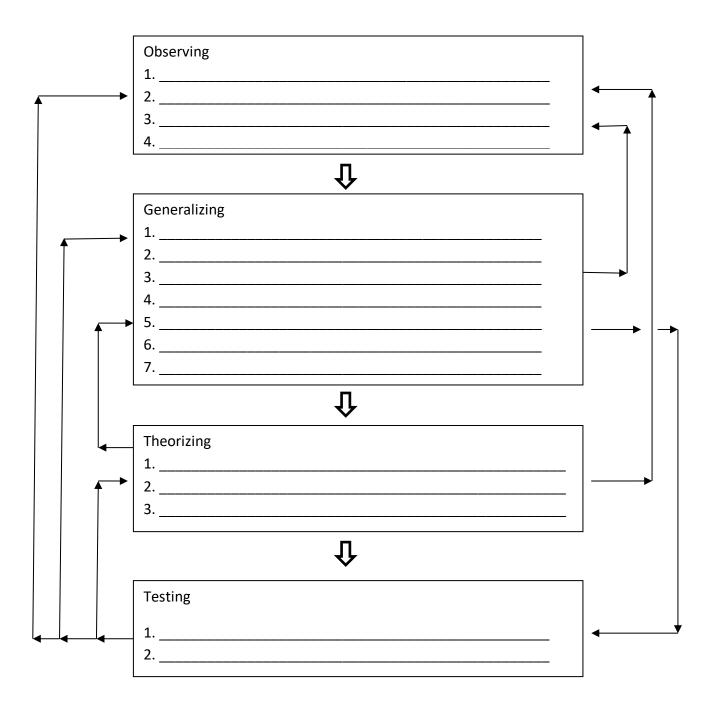
A **model** is a theory that can be visualized, and a **principle** or **law** is one idea within a more general theory. The word *law* can be misleading because it sounds so certain. As we will see, scientific ideas are never absolutely certain.

"Note that a theory is a well-confirmed framework of ideas. It's a misconception to think that a scientific theory is mere guesswork, as nonscientists occasionally do when they refer to some idea as 'only a theory'. Some people who disliked Copernican theory [heliocentric system] argued that it was a 'mere theory' that need not be taken seriously. Today, people who dislike the theory of biological evolution attack it on similar grounds. Theories - well-confirmed explanations of what we observe – are what science is all about and are as certain as any idea can be in science.

"The correct word for a reasonable but unconfirmed scientific suggestion (or guess) is **hypothesis**. For example, Kepler's first unconfirmed suggestion that the planets might move in elliptical orbits was a hypothesis. Once the data of Brahe and others confirmed Kepler's suggestion, elliptical orbits took on the status of theory rather than mere hypothesis."

- Dr. Art Hobson, "Physics -concepts and connections"

### **Scientific Method Flow Chart**



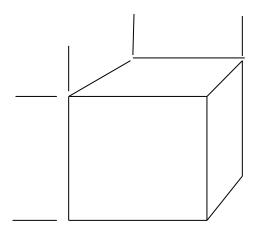
#### **IMPORTANT:**

"Communication": Common to \_\_\_\_\_\_ of the Scientific Method

# Key Points in the lingo and protocol of science and the Scientific Method

1. Theory:
2. Hypothesis:
3. Idealizations (Galileo):
4. Limiting the Scope of Inquiry (Galileo):
5. Quantitative methods (Galileo):
6. Creating a model:
7. Repeatable, predictable results/outcomes (Newton):
8. Fact-based rather than authority-based knowledge:

**Volume:** 



Volume = length x height x width

Therefore, \_\_\_\_\_ M³ is equivalent to \_\_\_\_\_ or \_\_\_\_ cm³

Additionally:

Using "factor-labeling" conversion method to determine # mL in 1 M³:

$$1 \text{ M}^3 \times 10^3 \text{ Liters } \times 10^3 \text{ milliliters} = \underline{\qquad} \text{ milliliters}$$

$$1 \qquad 1 \text{ M}^3 \qquad 1 \text{ Liter}$$

Since 1 M³ equals \_\_\_\_\_ mL,

$$_{\rm cm}^{\rm 3}$$
 =  $_{\rm mL}$ ; a cm<sup>3</sup> is also referred to as a " $_{\rm m}$ "

### **Mass Density Calculations**

Mass Density (D<sub>m</sub>) ( also referred to simply as "density")

is measured in \_\_\_\_\_ per \_\_\_\_ or \_\_\_/\_\_\_

### Example 1:

Determine the D<sub>m</sub> of a 67.5 gram sample of material with a volume of 30 cm<sup>3</sup>

Solution:

Use factor-labeling to convert grams/cm³ to Kilograms/M³

1. Restate the raw data as fraction:

$$\frac{67.5\ grams}{30\ cm^3}$$

2. Add conversion factors for cancellation:

$$\frac{67.5 \frac{grams}{30 \text{ cm}^3} \text{ X}}{30 \text{ cm}^3} \text{ X} \frac{1 \text{ Kg}}{10^3 \text{ grams}} \text{ X} \frac{10^6 \text{ } 10^3 \text{ cm}^3}{1 \text{ m}^3}$$

3. Restate with remaining terms and perform necessary calculations:

$$\frac{67.5 \ x \ 10^3 Kg}{30 \ m^3} = \underline{\text{Kg/m}^3}$$

NOTE:

Determining the  $D_{m}$  of a material can serve as an indicator of the chemical identity of the material.

### Example 2:

### Predicting the mass of a sample of known material

Given a 50 cm<sup>3</sup> sample of lead, predict the mass

**Solution**: Set up a proportionality equation using the known D<sub>m</sub> of lead

Step 1:

State the known D<sub>m</sub> of lead

$$\frac{11.3 \times 10^3 Kg}{m^3}$$

Step 2:

Set up as equivalent to given sample

$$\frac{11.3 \times 10^3 Kg}{m^3} = \frac{x g}{50 cm^3}$$

Step 3:

Convert all quantities to like terms (grams, cm<sup>3</sup>, since this is a small sample)

$$\frac{11.3 \times 10^6 g}{10^6 cm^3} = \frac{x g}{50 cm^3}$$

Step 4: Cross-multiply

$$\frac{11.3 \times 10^6 g}{10^6 cm^3} = 50 cm^3$$

Review: Cross-multiplying

-multiply
$$\frac{11.3 \times 10^{6} g}{10^{6} cm^{3}} \stackrel{=}{=} \frac{x g}{50 cm^{3}}$$

$$\frac{a}{b} = \frac{c}{d}$$

$$0 \times d = b \times c$$

$$0 \times d = d \times d$$

$$0 \times d = d \times$$

Now the equation becomes:

\_\_\_\_\_x \_\_\_\_= \_\_\_\_x \_\_\_\_

After canceling like terms, the equation becomes

\_\_\_\_\_ x \_\_\_\_ = \_\_\_\_

Thus, a 50 cm<sup>3</sup> sample of lead has a mass of \_\_\_\_\_ grams

# Neutron Stars: The ultimate in mass density:

If a star has sufficient	mass (that is to say, 8 to	ວ 20 times more tha	n our own
Sun)			
when it goes	, the atoms	of the remaining m	naterial in
	art by the extreme		
	ring this process electro		
	Since the volume of a i	-	
	e-empty volume is now		
	that a teaspoon of this		
	, or the same as	_	
000			岁步
			الله الله
4 4			4
000			
			4
由由由			

### Self check:

Determine the D <sub>m</sub> of a 273 gram sample of material with a volume of 35 mL	answer:
Determine the mass of a 95 cm <sup>3</sup> sample of iron	answer:
Determine the mass of a 95 cm <sup>3</sup> sample of iron	answer:
Determine the mass of a 95 cm <sup>3</sup> sample of iron	answer:
Determine the mass of a 95 cm <sup>3</sup> sample of iron	answer:
Determine the mass of a 95 cm <sup>3</sup> sample of iron	answer:
Determine the mass of a 95 cm <sup>3</sup> sample of iron	answer:
Determine the mass of a 95 cm <sup>3</sup> sample of iron	answer:
Determine the mass of a 95 cm <sup>3</sup> sample of iron	answer:
Determine the mass of a 95 cm <sup>3</sup> sample of iron	answer:
Determine the mass of a 95 cm <sup>3</sup> sample of iron	answer:

# **Vectors:**

Two types of measurement used in Physics; they are			
1	measures		
	b. Indicate	only	
2	measures		
	a. Indicate	and	
Examples:			
	Scalar	Vector	
Since vector	measures include the comp	onent of	then that
component	must be taken into considera	ation during	·

# Example 1: Two bungee cords pulling in opposite directions:

Example 1a:		
	• Net force:	
Evample 1h.		<del></del>
Example 1b:		
	•	
	Net force:	
Example 2: Two bungee	e cords pulling in the	same direction:
•		
	Net force:	
Note:		
At this point you should	I see the	of net forces, or
the "	-	n

				_	_
Example 3: Two	hungaa carde	nulling at a 0	Mo angla ral	ativa ta ana	anothar
LXaiiibie 5. IWU	ט טעווצפפ נטועא	Dullille at a 3	ואו מווצופ ופו	ative to one	anome

lacktriangle

Solution: Use "\_\_\_\_\_\_ to \_\_\_\_\_ " schematic

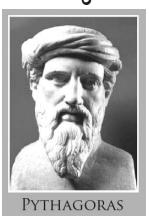
The sum of these two vectors is called a "\_\_\_\_\_\_"

### BUT,

The Pythagorean Theorem will solve \_\_\_\_\_\_ only

What about direction?

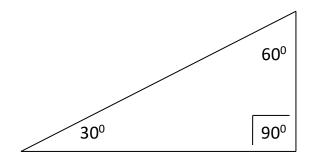




### **Basic Right-Angle Trigonometry**

(Invented by	ογ	)

Given: a 30-60-90 triangle



One of the unique characteristics of a 30-60-90 triangle is that

the side		_ the 30° angle	
is always _	<del>-</del>	the length of the	·

In other words, given the \_\_\_\_ angle,

the of the	side over the
------------	---------------

will always be \_\_\_\_\_\_, or \_\_\_\_\_.

In Right-Angle Trigonometry, this ratio is called

the \_\_\_\_\_ of the \_\_\_\_

Thus, we can state that the " $\_$ \_\_\_ of 30° is  $\_$ \_\_\_ "

### Trig on a calculator:

Depending on what model calculator you are using, you will do trig functions in one of two ways. We will use the **Sine (sin) of 30**° as an example.

### Method 1:

- 1. Hit sin
- 2. enter "30"
- 3. Hit "="
- 4. Your answer should be "0.5."
- 5. **If not**, you're probably in "radian mode" and using a graphing calculator.
- 6. Go to mode (you may need to use "shift" or "2<sup>nd</sup>" to get there)
- 7. You should see a screen showing both "degree" and "radian."
- 8. Select "degree"
- 9. enter
- 10. clear
- 11. Repeat steps 1 -3, your answer should now be "0.5"

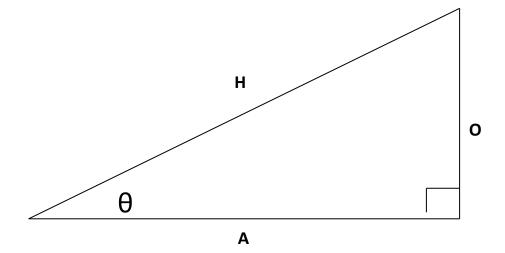
**Method 2** (more common on simpler, less expensive calculators):

- 1. Enter "30"
- 2. Hit sin
- 3. Your answer should be "0.5"

# Practice using other trig functions:

Cosine  $30^0 = .866$ 

Tangent  $30^{\circ} = .577$ 



θ ("Theta"):		
Hypotenuse:		
Opposite:		
Sine $\theta$ (sin $\theta$ ):		
the ratio of the	side over the	
<b>Cosine</b> $\theta$ (cos $\theta$ ):		
the ratio of the	side over the	
<b>Tangent 0</b> (tan $\theta$ )		
the ratio of the	side over the	side

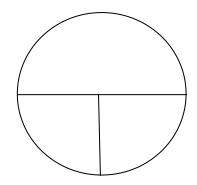
Stated more simply:

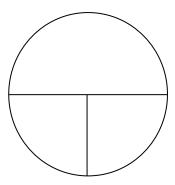
$$\sin\theta = \frac{O}{H}$$

$$\cos \theta = \frac{A}{H}$$

$$\tan \theta = \frac{O}{A}$$

Stated yet another way:





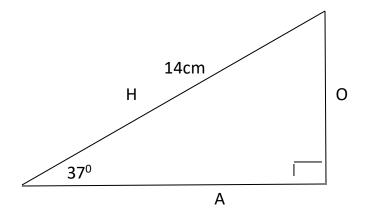
Sin 
$$\theta =$$

$$\cos \theta =$$

Tan 
$$\theta$$
 =

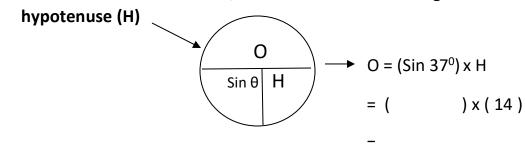
### **Using Trig**

Example 1.

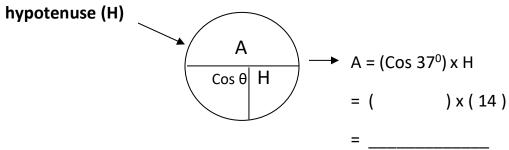


Determine the lengths of sides "O" and "A"

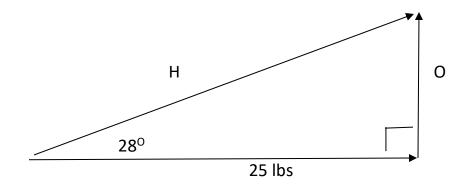
1. To determine "O" use "Sin", which uses the known angle and the known



2. To determine "A" use "Cos", which uses the known angle and the known

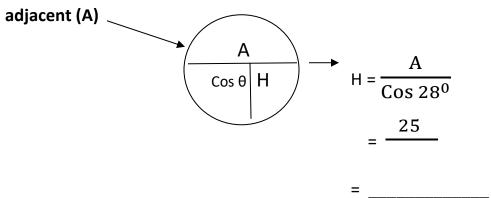


### Example 2.

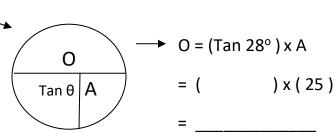


Determine the lengths of sides "H" and "O"

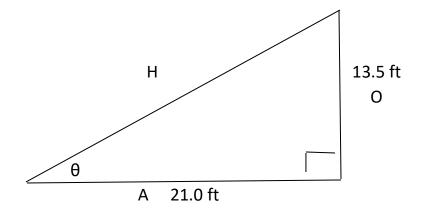
1. To determine "H" use "Cos", which uses the known angle and the known



2. To determine "O" use "Tan", which uses the known angle and the known adjacent (A)

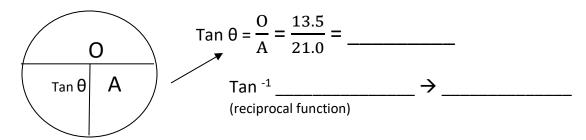


# Example 3:



Determine length of "H" and the value of " $\theta$  "

1. Begin by determining " $\theta$ ". Since "O" and "A" are known, use "Tan".



2. Determine "H" by using either **Sine** or **Cosine** 

$$H = \frac{O}{\sin \theta}$$

$$H = \frac{A}{\cos \theta}$$

# Using the **"reciprocal function"** on the calculator

Since we know that the **Sine of 30<sup>o</sup> is 0.5**, we'll start there.

### Method 1:

- 1. Hit "Shift" or " 2nd "
- 2. Hit sin
- 3. Enter ".5 "
- 4. Hit " = "
- 5. Your answer should be "30"

### Method 2 (for simpler, less expensive calculators):

- 1. Enter " .5 "
- 2. Hit "Shift" or " 2<sup>nd</sup> "
- 3. Hit sin
- 4. Your answer should be "30"

# Practice using other trig functions:

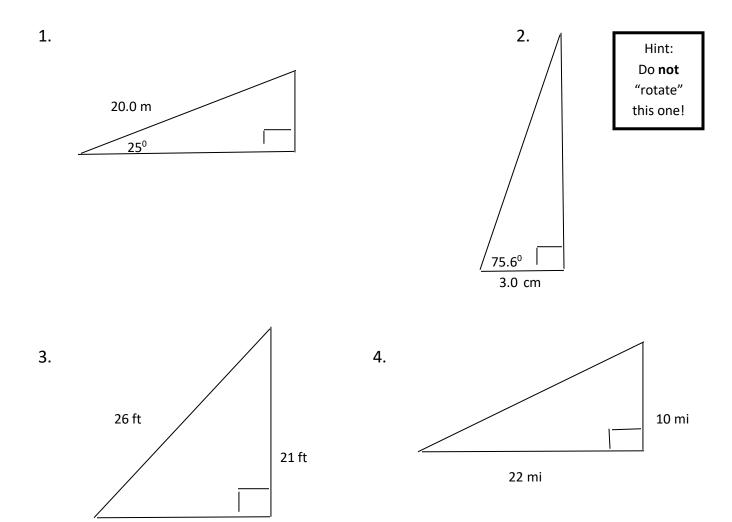
$$\cos^{-1}.866 \rightarrow 30^{0}$$

$$tan^{-1}.577 \rightarrow 30^{0}$$

# **Informal Lab: Practice problems**

**ASSIGNMENT**: Solve for unknown sides and angles using trigonometry **plus** additional instructions below:

- 1. Do **NOT** Pythagorean Theorem!
- 2. **DRAW** these in larger scale on a separate paper use a straight-edge if it helps. The idea is to get you used to drawing, **as Galileo recommends!**



### Practical principles:

1. The sine, cosine, and tangent of any/every angle between \_\_\_\_\_ and \_\_\_\_\_

is \_\_\_\_\_ to that angle alone.

2. Thus, if we know the sine, cosine and/or tangent of an angle, then we have

the means to \_\_\_\_\_ the original angle.

#### 3. NOTE:

This course will take the "old-school trig" approach in analyzing angles,

and therefore all angles ("vectors") will be evaluated as if their

measures are between \_\_\_\_\_ and \_\_\_\_\_ .

### Example 1:

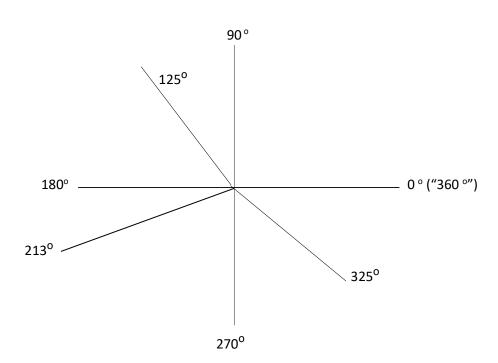
 $125^{\circ}$  will be evaluated as \_\_\_\_\_ (  $180^{\circ}$  -  $125^{\circ}$  ) (measure from x-axis)

Example 2:

213° will be evaluated as \_\_\_\_\_ (213° -180°) (measure from x-axis)

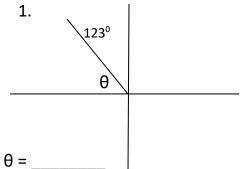
Example 3:

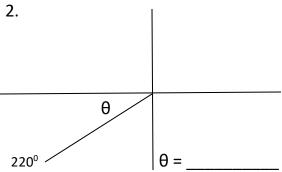
 $325^{\circ}$  will be evaluated as \_\_\_\_\_ ( $360^{\circ}$  -  $325^{\circ}$ ) (measure from x-axis)



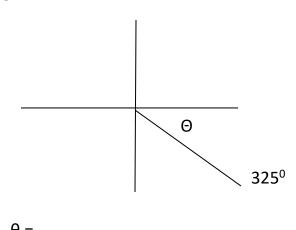
# Informal Lab: Practice evaluating angles

Hint: Always measure from the <u>closest horizontal</u> ("x - axis")

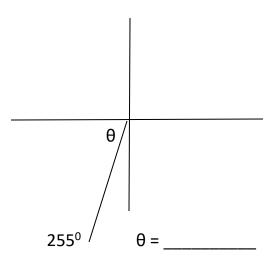




3.

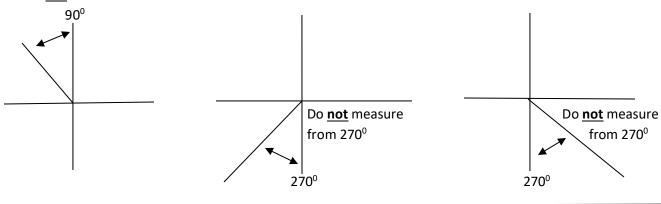


4.



#### **Common mistakes:**

Do **not** measure from



**ANSWERS: 1.** 57<sup>0</sup> (180 – 123) **2.** 40<sup>0</sup> (220-180) **3.** 35<sup>0</sup> (360-325)

**4.** 75<sup>0</sup> (255-180)

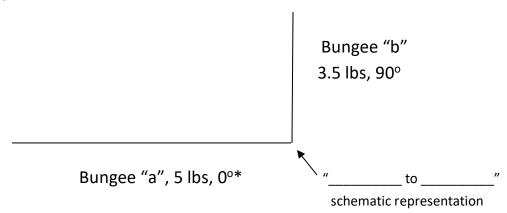
### Back to the beginning of this topic:

Given:

- 1. Two bungee cords attached at a common point.
- 2. Bungee "a" pulls with 5 pounds of force at 0°
- 3. Bungee "b" pulls with 3.5 pounds of force at 90°
- 4. What is the sum of the forces of the two bungee cords?

Solution:

Draw \*:



\* NOTE: Always draw \_\_\_\_\_\_ vector first beginning with the \_\_\_\_\_

- 1. Calculate the tangent of the unknown angle "\_\_\_\_": tan = —— = \_\_\_\_\_
- 2. tan<sup>-1</sup> \_\_\_\_\_
- 3. Calculate the hypotenuse ( or "\_\_\_\_\_\_")

Using trig, we know that H =  $\frac{O}{\sin \theta}$  and/or  $\frac{A}{\cos \theta}$ 

Selecting the first trig formula,

H = \_\_\_\_\_ = (solution)

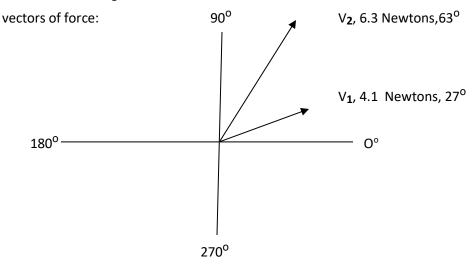
Thus, the sum ( \_\_\_\_\_\_ ) of the forces of the two bungee

cords is \_\_\_\_\_ at \_\_\_\_ degrees

### **Adding Vectors:**

#### Example 1:

Given the following



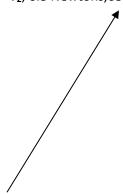
Determine the sum of  $V_1 + V_2$ 

#### Solution:

- ① Draw each vector individually, and label accordingly
- ② Add component vectors and label accordingly
- $\ensuremath{\Im}$  Use trig to solve for components
- Add components
- ⑤ Construct new vector ("resultant") using component sums
- © Use trig to evaluate resultant

V<sub>1</sub>, 4.1 Newtons, 27°



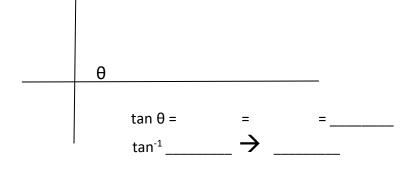


Construct/draw in order:

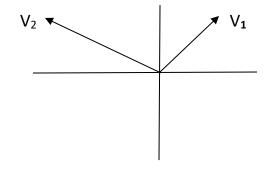
A total

O total

Resultant

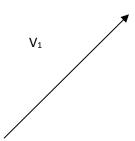


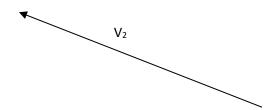
# Example 2:

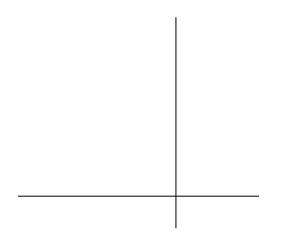


V<sub>1</sub>: 48°, 50 meters/sec V<sub>2</sub>: 147°, 75 meters/sec

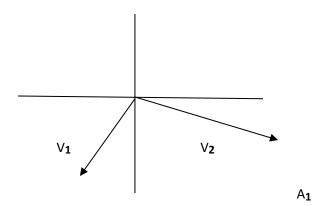
$A_1$	$O_1$	
A <sub>2</sub>	O <sub>2</sub>	
At	Ot	







Example 3:

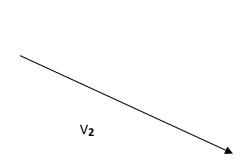


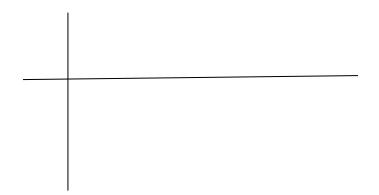
V<sub>1</sub>: 245°, 36 N V<sub>2</sub>: 337°, 68 N



A<sub>2</sub> O<sub>2</sub> A<sub>t</sub> O<sub>t</sub>

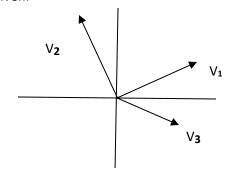
01





### Forces in Equilibrium

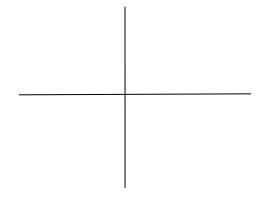
#### Given:



 $V_{\mathbf{1}}$ 

 $V_{\mathbf{2}}$ 

٧3



V<sub>1</sub>: 33°, 3.1 N

V<sub>2</sub>: 103<sup>o</sup>, 2.0 N

V<sub>3</sub>: 338°, 1.6 N

#### **Calculations:**

A <sub>1</sub>	O <sub>1</sub>	
A <sub>2</sub>	O <b>2</b>	
A <sub>3</sub>	O <sub>3</sub>	
At	Ot	

Resultant calculations:

Tan θ\_\_\_\_\_ = \_\_\_\_\_

$$H = \frac{O}{\sin \theta} = \frac{O}{O}$$

Calculate the vector that will cancel the resultant Equilibrium Vector (  $V_{eq}$  )calculations

Note: What is the sum of all component force vectors in a system in equilibrium?

# **MOTION**

1. \	/elocity:	
	a measurement	
	b and	
	c over ( )	
	d. Measured in:	
	1 per (/) (British)	
	2 per (/) (Metric)	
2.	Average Velocity (V <sub>avg</sub> )	
	Averages in velocity over a given period of time Example: Driving from Portland to Boston	ž.
3.	Uniform Velocity:	
	Velocity that does not (Example: ""	

\_

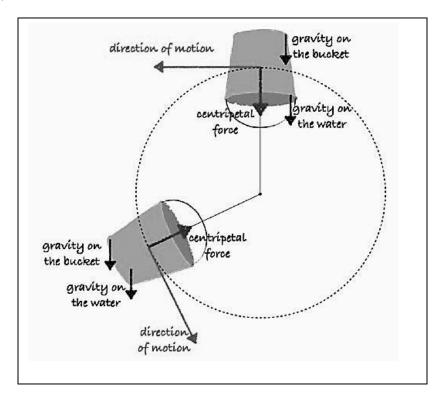
4.	Accel	leration	•
/I		Pration	١
┰.		Clation	

a		_ measur	rement		
b. A		_ in	* over	r	
	*( "V∆" or '	u 	V")		
	( "Δ" = "_		_")		
c. Me	asured in:				
	1	per	per	(	/ ) (British)
	2	per	per	(	/) (Metric
•	"ft/sec² = f	eet per se	econd per seco second per sec	•	

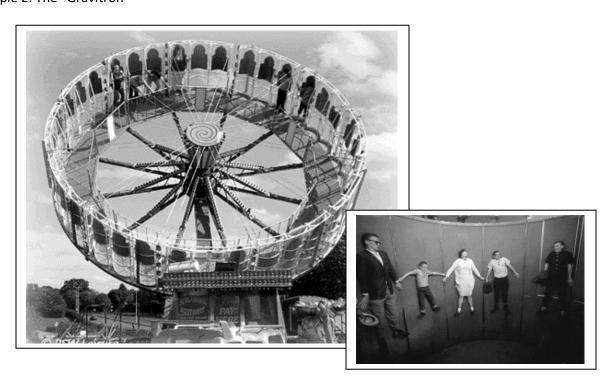
# Acceleration (continued):

1. The acceleration of gravity ( a <sub>g</sub> )	CAUTION:
On Earth:	Gravity is acceleration, <b>BUT</b>
a ft/sec² (British)	Not all acceleration is gravity!
bm/sec² (Metric)	
c. Thus, one "g" = or	
2. Key words:	2 72116 1
a. "Boost":	Law of Gravity Strictly Enforced!
b. "Retro-burn"	En In Bala
c. Negative g's	
3. Acceleration due to a change in direction	1:
Since acceleration is defined as	
a in ,	
and velocity is defined as	
and ,	
then a in re	sults in

Example 1: Beach Bucket



Example 2: The "Gravitron"



# **Critical Factors in Acceleration Calculations:**

	1. V <sub>i</sub> :				
	2. V <sub>f</sub> :				
	3. a:				
	4. s:				
	5. t:				
		•	" Thre	e out	a five ain't bad!"
		Given	any		of the above factors,
	the	remai	ning		factors may be calculated
		Ва	sic For	mulas	Used in Acceleration Problems:
Vi	V <sub>f</sub>	a	S	t	basic formula:
				<u> </u>	

### **Acceleration Formula Cheat Sheet**

$$t = \otimes$$

$$a = \frac{V_{f^2} - V_{i^2}}{2s}$$
  $s = \frac{V_{f^2} - V_{i^2}}{2a}$   $V_f = \sqrt{V_{i^2} + 2as}$   $V_i = \sqrt{2as - V_{f^2}}$ 

$$S = \frac{V_{f^2} - V_{i^2}}{2a}$$

$$V_f = \sqrt{V_{i^2} + 2as}$$

$$V_i = \sqrt{2as - V_{f^2}}$$

$$s = \otimes$$

$$V_f = at + V_i$$

$$V_f = at + V_i$$
  $V_i = V_f - at$   $a = \frac{V_f - V_i}{t}$   $t = \frac{V_f - V_i}{a}$ 

$$a = \frac{V_f - V_i}{t}$$

$$t = \frac{V_f - V_i}{a}$$

$$a = \otimes$$

$$s = .5(V_f + V_i)t$$

$$s = .5(V_f + V_i)t$$
  $t = \frac{s}{.5(V_f + V_i)}$   $V_f = \frac{s}{.5t} - V_i$   $V_i = \frac{s}{.5t} - V_f$ 

$$V_f = \frac{s}{.5t} - V_i$$

$$V_i = \frac{s}{.5t} - V_f$$

$$V_f = \otimes V_i = o$$

$$V_i = 0$$

$$s = .5at^2$$

$$s = .5at^2 a = \frac{s}{.5t^2} t = \sqrt{\frac{s}{.5a}}$$

$$t = \sqrt{\frac{s}{.5a}}$$

$$V_f = \otimes V_i \neq o$$

$$V_i \neq 0$$

$$s = V_i t + .5at^2$$

$$s = V_i t + .5at^2$$
  $t = \frac{-V_i \pm \sqrt{V_{i2} + 2ac}}{a}$   $v_i = \frac{.5at^2}{t}$   $a = \frac{s - v_i t}{.5t^2}$ 

$$v_i = \frac{.5at^2}{t}$$

$$a = \frac{s - v_i t}{.5t^2}$$

# Points to ponder

Given this formula:  $s = V_i t + \frac{1}{2} a t^2$ 

Solve for  $oldsymbol{t}$ 

$$(3) + (-s) + (-s)$$

$$\Im .5at^2 + V_i t + (-s) = 0$$

6

Change symbol		
from	to	

6

### Informal Lab: Working through acceleration problems

Example 1:

How long will it take an object to drop 4 feet?

1

2) (3

Step 1: Does this question involve gravity and /or acceleration? If so, then go to:

Step 2: Inventory

 $V_i$ 

 $V_f$ 

② **a** 

③ **S** 

① **t** 



Step 3: What is the question?

Step 4: Look for the "odd man out" ( $\otimes$ )

Step 5: Look for ⊗ on the Cheat Sheet

Step 6: Select formula corresponding to "?"

Step 7: Insert correct values in formula and solve

- Be sure to use correct standard units! Convert if necessary.

Example 2:
(Part 1)
A rock is <u>dropped</u> from a bridge. It takes <u>1.35 seconds</u> for the rock to strike the water
below. How high (in ft) is the bridge above the water?
$V_i$
$V_{f}$
a
S
t
l de la companya de
(Part 2)
How fast is the rock travelling at impact?
$V_i$
$V_{f}$
a
S
t

### Example 3.

A ball is thrown straight down from a cliff. The velocity of the ball as it leaves the thrower's hand is 60 ft/sec. How far will the ball have travelled after 2 sec.?

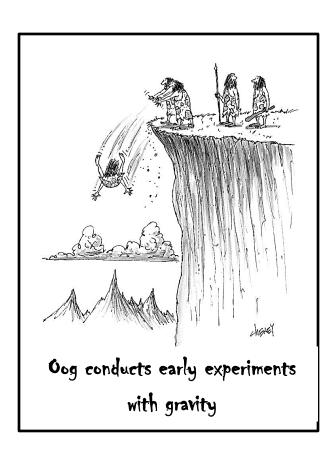
 $V_{i}$ 

 $V_{\mathsf{f}}$ 

a

S

t



Example 4.

A rocket boosts from the launch pad at 48 ft/sec<sup>2</sup>. How high is the rocket after 5 sec.?



Example 5.

A car goes from 55MPH to 70 MPH in 10 sec. What is its rate of acceleration? (Hint: convert to standard units **first**)

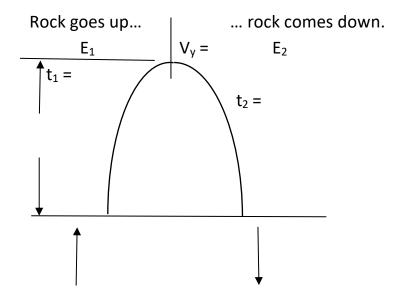
### Example 6.

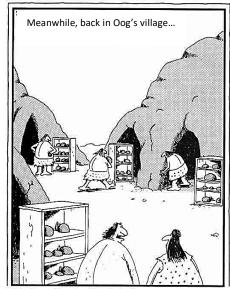
An aircraft with a landing speed of 180 MPH lands on an aircraft carrier by catching the arresting wire and coming to a complete stop in 2 sec. How many G's does the pilot experience? (Be sure to convert to correct units first!)











"You know, I used to like this hobby. ... But shoot! Seems like everybody's got a rock collection."

Q1: What is the initial velocity (V <sub>i</sub> )of the rock going up? V <sub>i</sub> V <sub>f</sub> a s t		Q2: How long does it take the reach max height?  Vi  Vf a s t	ne rock to
	V <sub>i</sub> =		t =
Q 3: How long does it take come back down?  Vi  Vf  a  s t	e the rock to	Q4: What is the final velocit at the return point?  V <sub>i</sub> V <sub>f</sub> a  s  t	y of the rock
	t =		V <sub>f</sub> =

### **Informal Lab Problems:**

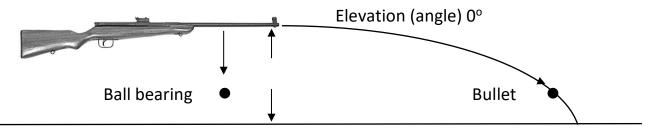
- 1. A bullet is fired vertically with an initial velocity 0f 250 m/sec. Discounting air resistance,
  - a. How high does it go?
  - b. How long does it take to reach max height?



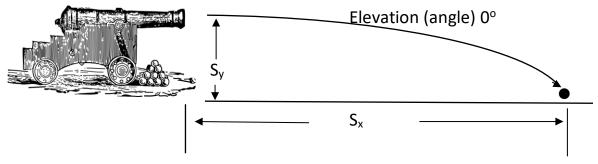
- 2. A bullet is fire vertically and reaches a max height of 700 ft Discounting air resistance,
  - a. What is its initial velocity?
  - b. How long does it take to reach max height?

#### **Kinematics: Motion in Two Dimensions**

#### Example 1:



### Example 2:



### Using a level shot to determine Muzzle Velocity (V<sub>muzzle</sub>)

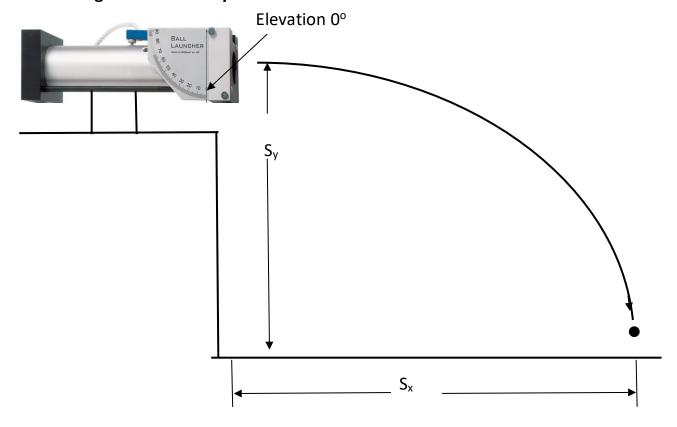
Theory:

- 1. Definition of velocity: (---)
- 2. V<sub>x</sub> is stipulated as \_\_\_\_\_\_ ("idealization")
- 3.  $V_{\text{muzzle}}$  is set at  $0^{\circ}$  elevation and is therefore \_\_\_\_\_\_ to  $V_x$
- 4. "S" is measured as \_\_\_\_\_\_
- 5. "t" is calculated using \_\_\_\_\_

## **Calculations/measurements:**

- 1. S<sub>x</sub> \_\_\_\_\_
- 2. S<sub>y</sub>\_\_\_\_\_
- 3. t=
- 4.  $V_{\text{muzzle}} =$

# **Determining Muzzle Velocity**



$$S_y =$$
 (as measured)

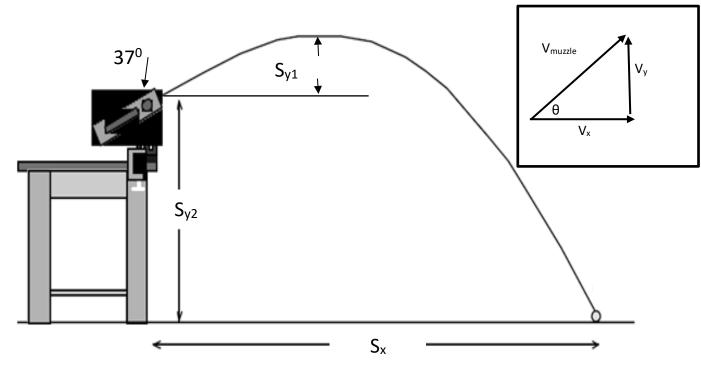
$$S_{x=}$$
 (as measured)

$$t = \sqrt{\frac{S_y}{.5a}} =$$

$$V_{x} = \frac{S_{x}}{t} =$$
 —  $V_{muzzle}$ 

(only when elevation is set at 0°)

# Predicting range of angled shot based on known $V_{\rm m}$

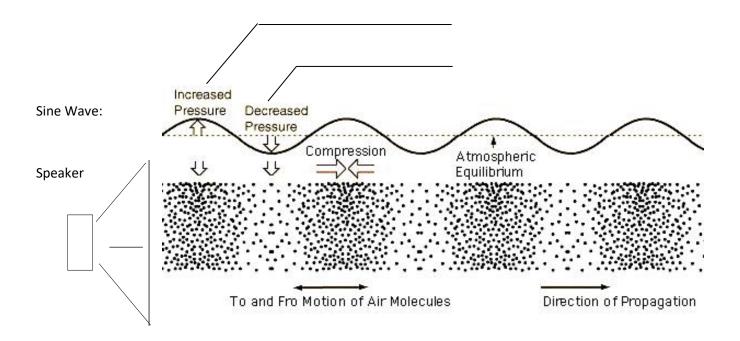


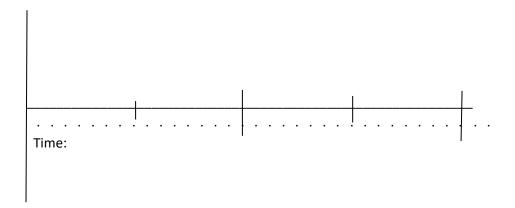
### OBJECTIVE: Predict S<sub>x</sub> given known V<sub>muzzle</sub>

	0			
1. Proposition:	$S_x = V_x x$	$t_{total,} \longrightarrow$	$(t_{total} = [(t_1) +$	+ (t <sub>2</sub> )]
	$\overline{\downarrow}$			
	= (	) × [(	)+(	)] =

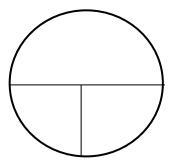
3.	$V_x =$			4.	V <sub>y</sub> :

# Sound:





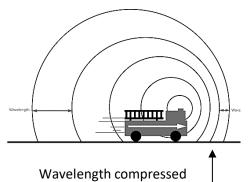
Frequency, wavelength, velocity:



#### The role of the medium in a mechanical wave

The medium determines \_\_\_\_\_\_ of a wave

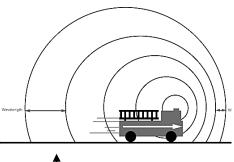
The Doppler Effect:



relative to observer



observer



Wavelength expanded relative to observer

Longer \_\_\_\_\_ = lower \_\_\_\_\_ Shorter \_\_\_\_\_ = higher \_\_\_\_\_

Doppler equation:  $f' = f\left(\frac{V}{V \pm V_s}\right)$ 

#### Where:

v<sub>s</sub> = Velocity of the Source

v = Velocity of wave

f = Real frequency

f' = Apparent frequency

**Equation to determine velocity of source:** 

$$V_s = \frac{V(f'-f)}{f'}$$

#### Doppler Effect Real World Example:

A sonar analyst detects an underwater sound at a frequency of 319.63 HZ. He knows from prior intelligence that sound is actually propagated at 318.00 hz.

- 1. Is the sound source approaching or receding?
- 2. What is the speed of the source in Knots (nautical miles per hour)?

#### Data:

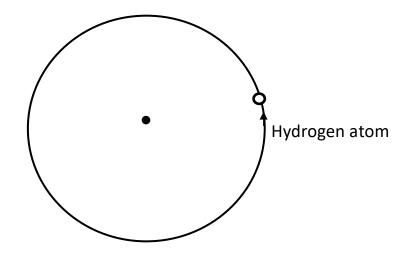
- 1. Speed of sound in water ~ 4900 ft/sec
- 2. 1 Nautical mile  $\sim 6000$  ft.

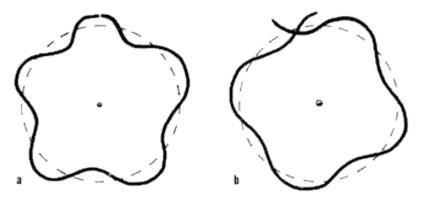


Soviet submarine with US Navy P-3 Orion anti-submarine surveillance aircraft (My old alma mater – Patrol Squadron Eight!)

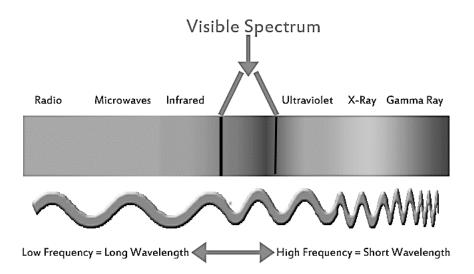
# Structure of the atom and the nature of light

### Recall:



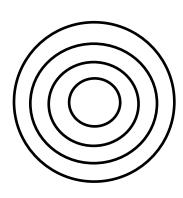


De Broglie Wavelength



Wile E. Coyote's Last Hurrah!





#### **Hubble Law**

**Hubble's law** or **Hubble—Lemaître's law** is the name for the observation that:

- 1. All objects observed in deep space (extragalactic space, ~ 10 Mpc or more) have a doppler shift-measured velocity relative to Earth, and to each other;
- 2. The doppler-shift-measured velocity of galaxies moving away from Earth, is proportional to their distance from the Earth and all other interstellar bodies.

In effect, the space-time volume of the observable universe is expanding and Hubble's law is the direct physical observation of this. It is the basis for believing in the **expansion of the universe** and is evidence often cited in support of the Big Bang model.

Although widely attributed to Edwin Hubble, the law was first derived from the General Relativity equations by Georges Lemaître in a 1927 article. There he proposed that the Universe is expanding, and suggested a value for the rate of expansion, now called the **Hubble constant**. Two years later Edwin Hubble confirmed the existence of that law and determined a more accurate value for the constant that now bears his name. The recession velocity of the objects was inferred from their redshifts, many measured earlier by Vesto Slipher in 1917 and related to velocity by him.

The law is often expressed by the equation  $v = H_0D$ , with  $H_0$  the constant of proportionality (the **Hubble constant**) between the "proper distance" D to a galaxy and its velocity v (see *Uses of the proper distance*).  $H_0$  is usually quoted in (km/s)/Mpc, which gives the speed in km/s of a galaxy 1 megaparsec (3.09×10<sup>19</sup> km) away. The reciprocal of  $H_0$  is the Hubble time.

**Hubble law**:  $V = H_oD$ 

Where:

V = velocity in Km/sec

Ho = Hubble Constant =  $\frac{71 \text{ Km/sec}}{\text{Mpc}}$ 

D = distance in parsecs (pc)

1 parsec (pc) = 3.26 LY

# Example:

Astronomers observe a galaxy 7 billion light years away.

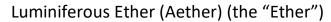
- 1. How fast is the galaxy moving away from us?
- 2. How long has it been travelling?

### Michelson - Morley Experiment

Albert Michelson

1. \_\_\_\_\_

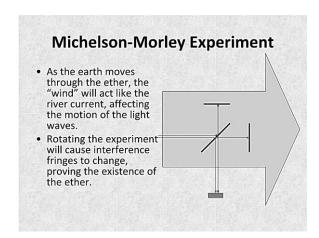
2.

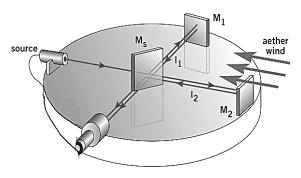


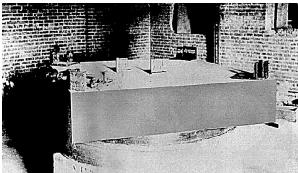
1. \_\_\_\_\_

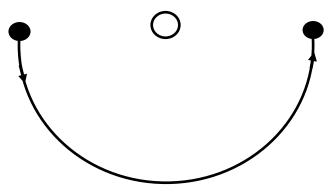
2.





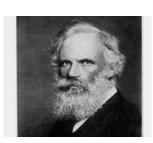


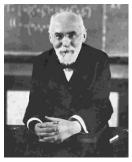




George Fitzgerald \_\_\_\_\_\_

Hendrik Lorentz : \_\_\_\_\_



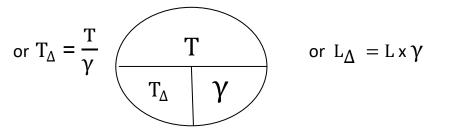


**Lorentz Factor:** 
$$\sqrt{1-rac{V^2}{c^2}}$$
 or  $\gamma$  ("Gamma")

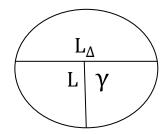
Where

Time: 
$$T_{\Delta} = \frac{T}{\sqrt{1 - \frac{V^2}{c^2}}}$$
 Length:  $L = L_{\Delta} \times \sqrt{1 - \frac{V^2}{c^2}}$ 

Length: 
$$L = L_{\Delta} \times \sqrt{1 - \frac{V}{c^2}}$$



or 
$$\mathsf{L}_\Delta \, = \mathsf{L}\,\mathsf{x}\,\gamma$$



# **Relativity Toolbox**

Where:		
$T_{\Delta} =$		
T =		
C =		
V =		
$L_{\Delta} = $		
L =		
Relativistic Velocities ()		
Non-Relativistic Velocities (	)	
"Gamma" ( $\gamma$ ) is the factor that allows us $\gamma$		
evident at		
velocities.		

### **Einstein's Two Postulates of Special Relativity:**

1. The laws of physics	 	
2. The speed of light		

### **Quotes by Albert Einstein:**

### On Relativity:

"When you are courting a nice girl, an hour seems like a second. When you sit on a red - hot cinder, a second seems like an hour. That's relativity."

#### On virtue:

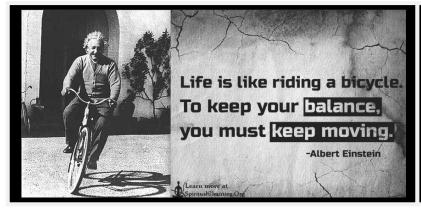
"As far as I'm concerned, I prefer silent vice to ostentatious virtue."

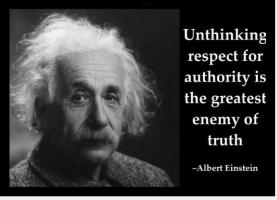
# On traffic safety:

"Any man who can drive safely while kissing a pretty girl is simply not giving the kiss the attention it deserves."

#### On nationalism:

"Nationalism is an infantile disease. It is the measles of mankind."

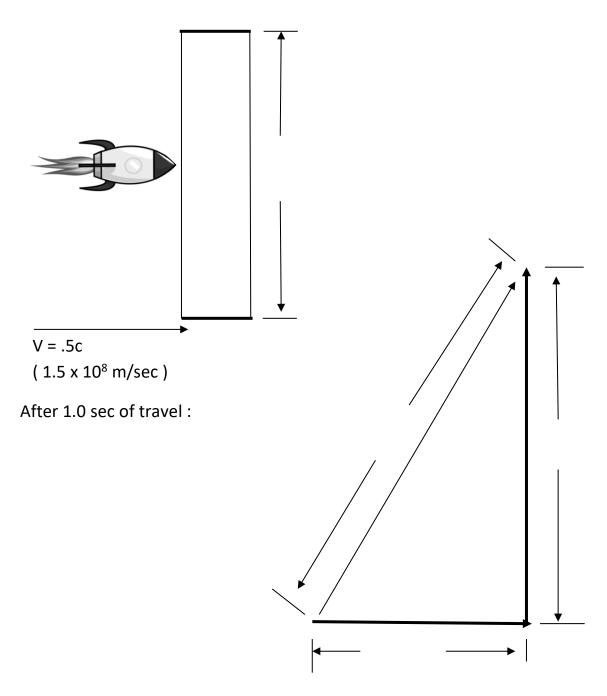




To understand why Relativity is necessary we have to look at the practical problems resulting from a Cosmic Speed Limit (The speed of light: "c")

(C= 186,000 mi/sec, 300,000 km/sec, and/or 3.0 x 10<sup>8</sup> m/sec)

We'll start with a ridiculous imaginary clock:

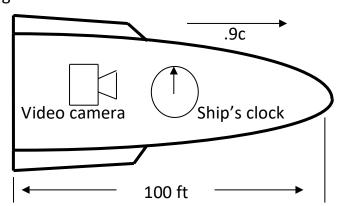


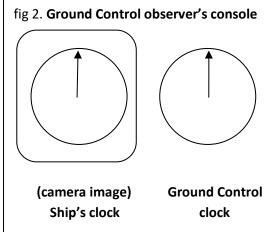
#### **Relativity Example 1.**

A spacecraft passes NASA Ground Control at .9c.

A video camera monitors the clock inside the cabin and transmits the image to an observer in Ground Control. The observer has his own clock adjacent to the console video screen displaying the shipboard clock.

fig. 1





**Question 1:** The Ground Controller observes the image of the Ship's clock second hand as it completes 1 rotation (60 sec). How much time has elapsed on the Ground Control clock?

**Step 1:** Calculate "Gamma"  $(\gamma)$ 

$$\gamma = \sqrt{1 - \frac{V^2}{c^2}}$$

Step 2: Solve for 
$$T_{\Delta}$$
 
$$T_{\Delta} = \frac{T}{\gamma}$$

Question 2: What is the length of the spacecraft from the perspective of the observer?

$$\mathsf{L} = \mathsf{L}_{\Delta} \, \mathsf{x} \, \gamma$$

#### Relativity and the Muon

Evidence supporting Einstein's theory of Special Relativity is found in the analysis of the behavior of muons.

Muons are subatomic particles that are created in Earth's upper atmosphere when cosmic rays (typically protons) collide with the nuclei of air molecules; muons have a velocity of .998c and a "life span" of **2.2 x 10**<sup>-6</sup> **seconds** (at rest), after which they disintegrate into other particles.

Scientists conducted an experiment in which they detected the presence of muons at the top of Mount Washington, New Hampshire.

After recording their results, they then moved their detection equipment to a New England beach ("sea level").

Given the altitude of Mt. Washington (approximately 2000 meters), and the velocity (V) and "life span" (T) of muons, (and discounting the effects of Relativity) there should have been no muons detected at sea level, since:

(V) x (T) = (Distance)  

$$\downarrow$$
 (Distance)  
(.998c) x (2.2 x 10<sup>-6</sup>) = (2.994 x 10<sup>8</sup> m/sec) (2.2 x 10<sup>-6</sup> sec) = 658.68 meters

In other words, <u>according to classical Newtonian principles</u> the muons should have disintegrated a little over a third of the distance down from the top of the mountain.

Yet, when the detection equipment was activated at sea level, muons were clearly and abundantly present!

#### Solution:

1. Calculate "Gamma" for .998c

2.	Calculate T <sub>A</sub>
ے.	Carcarate 14

3. Calculate  $L_{\Delta}$  from the perspective of the muon:

## Famous quotes by baseball legend and American philosopher Yogi Berra:

#### On Relativistic Time:

"This is the earliest I've ever been late!"

## **On Quantum Physics:**

"When you come to a fork in the road, take it."

### On the Abstract Mathematics:

"Baseball is ninety percent mental and the other half is physical."



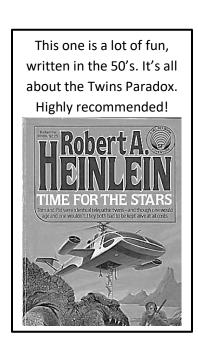
#### **The Twins Paradox**

One of pair of identical twins is selected to be a crew member of a deep-space expedition to a star eleven light-years distant.

The other twin will remain on Earth.

The vessel will travel at .998c

Discounting the time spent exploring the star system, determine the ages of each twin upon the vessel's return to Earth



## **Gamma Chart For Relativistic Velocities**

V	√ <sup>2</sup>	1-v <sup>2</sup>	√( 1-v <sup>2</sup> ) (" <b>γ</b> ")
.9c (.1 or one-tenth under "c")	.81	.19	.44
.99c (.01 or one-hundredth under "c")	.980	.02	.14
.999c (.001 or one-thousandth under "c")	.998	.002	.045
.9999c (.0001 or one-ten thousandth under "c")	.9998	.0002	.014
.99999c (.00001 or one-hundred thousandth under"c")	.99998	.00002	.0045
.999999c (.000001 or one-millionth under "c")	.999998	.000002	.0014
.9999999c (.0000001 or one-ten millionth under "c")	.9999998	.0000002	.00045
.99999999c (.00000001 or one-hundred millionth under "c")	.99999998	.00000002	.00014
.99999999c (.00000001 or one-billionth under "c")	.999999998	.000000002	.000045
.999999999c (.000000001 or one-ten billionth under "c")	.999999998	.0000000002	.000014

Further Problems with Relativistic Travel (example 1):

A crew of astronauts leaves Earth to explore deep space.

#### Given:

- 1. From the crew's perspective, they will experience one year of shipboard time travelling within a billionth of "c".
- 2. "Gamma" for their velocity is 0.00001 (See chart on previous page)

Determine how much time will have elapsed on Earth when they return.

Further Practical Problems with Relativistic Velocity (Example 2)

Given: A space vessel traveling at .9c collides with an small object with a mass of grain of salt, approximately  $5.86 \times 10^{-8} \, \text{Kg}$ 

How much kinetic energy ( KE ) is released at impact?

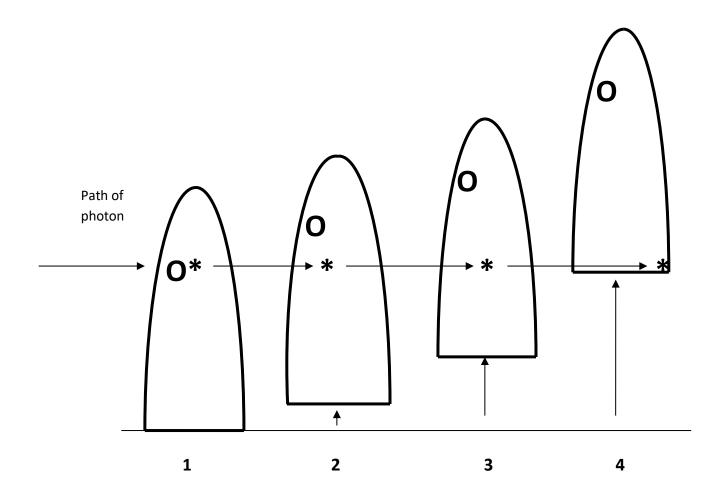
(Comparison: 1 ton of TNT =  $4.2 \times 10^9$  Joules)

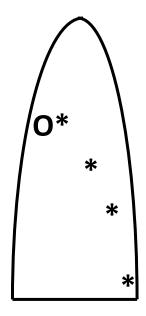
Given: A space vessel traveling at .9c collides with an small object with a mass of 2.5 grams ( roughly the mass of a penny )

How much kinetic energy ( KE ) is released at impact?

Further Practical Problems with Relativistic Velocity (Example 3)

# The effects of acceleration on the path of a photon





Path of photon relative to spacecraft

# An Einstein "Thought Experiment"

If the Sun was to suddenly vanish, would the Earth break from its orbit at the instance of the Sun's disappearance?

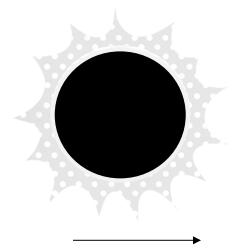
Newton's View

Einstein's view:

Proof of gravity affecting light during solar eclipse:









## **Another Thought Experiment:**

Escape Velocity:

Formula for Escape Velocity:

$$V_{esc} = \sqrt{\frac{2GM}{r}}$$

Calculate Escape Velocity (  $V_{\text{esc}}$  ) for Earth

Data:

Radius of Earth: 6378 Km Mass of Earth: 6.0 x 10<sup>24</sup> Kg

Universal Gravitational Constant (G):

6.672 x 10<sup>-11</sup>

Now let's super- shrink the Earth and reduce the radius to 7.8 mm ( $7.8 \times 10^{-3}$  m) and calculate the new escape velocity.



### The Most Famous Equation in the World:

$$E = mc^2$$

To get a handle on this, let's first take a look at a lesser known version:

$$E = mc^2$$

Where:

E = "Binding Energy" m = "mass defect"

 $C^2$  = speed of light squared  $(3.0 \times 10^8)^2$ 

Data:

Mass of a proton =1.67262 x 10  $^{-27}$  Kg Mass of a neutron = 1.67493 x 10 $^{-27}$  Kg Mass of an electron = 9.1094 x 10  $^{-31}$  Kg

We'll start by constructing a Helium atom and predicting its mass based on the known masses of its constituent parts.

Remember, a Helium atom contains 2 protons, 2 neutrons, and 2 electrons

<sup>4</sup>H compared to <sup>235</sup><sub>92</sub>U

Top#\_\_\_\_\_

Botton#\_\_\_\_

2 protons	_ Kg	
+ 2 neutrons	Kg	
Predicted total =	Kg	
Actual total = <b>6.6463</b> X 10 <sup>-27</sup> Kg		
Difference:	Kg	
(Missing mass or "		″\

E = mc<sup>2</sup>

= \_\_\_\_\_x

= \_\_\_\_x

Ioules

Now compare the mass – energy conversion factor:

Original mass		
Resulting energy		

Note the exponential difference

# Finally, **E** =mc<sup>2</sup>

An alternate way to read the formula:

"There is an equivalence between mass and energy, with a conversion factor that is the square of the speed of light"

### **Question:**

How much TOTAL energy is contained in 1 Kilogram of material (like the Laboratory Rock)?

BOOM!



## Epilogue: Where do we go from here?

The Four Fundamental Forces: B.B. 0 sec - 10<sup>-43</sup> sec Described by: Conflict: The existence of String Theory – possible solution? All "matter" is Original model called for the existence of Problems developed because of mathematical Anomalies resolved by \_\_\_\_\_\_ Strength of String Theory: Weakness of String Theory:

