

# CHEM 2410 Fall 2018 – Mid-Term Exam 2 10-24-18

Time: 5:30pm - 6:30pm

Student Name:	
Student Number:	Prof. Andreana WO 1205

1. What would be the **major** organic product of the following  $E_2$  reaction?

$$\begin{array}{c|c} & & K^{+}\text{-}OC(CH_{3})_{3} \\ \hline & & \\ \hline & & HOC(CH_{3})_{3} \end{array} ?$$

A)

B)

C)

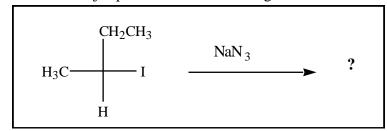
D)

E)

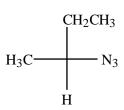
2. Which set of reagents will best accomplish the following reaction?

- A) Br<sub>2</sub>, acetone
- B)  $H_2SO_4$ ,  $H_2O$
- C) Br<sub>2</sub>, hv
- D) NaOEt, DMSO

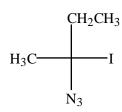
3. Predict the major product of the following  $S_{\rm N}2$  reaction:



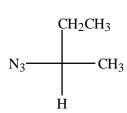
A)



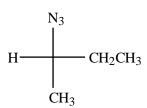
B)



C)

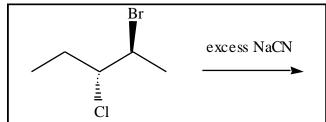


D)



E) none of the above

4. What is the correct stereochemistry of the product of the following  $S_N2$  reaction (Hint: Two  ${}^{\text{-}}\!CN$  nucleophiles react – one at each chiral center):



- A) 3*R*,4*S*
- B) 2*S*,3*R*
- C) 2*R*,3*S*
- D) 2R,3R
- E) 3R,4R
- 5. Predict the product of the following  $E_2$  reaction.

$$\begin{array}{c|c} H & H \\ Ph^{II} & Ph \\ H_3C & Br \end{array}$$

A)

$$H$$
 $Ph$ 
 $H_3C$ 
 $N(i-Pr)_2$ 

B)

$$\begin{array}{ccc} & & & \text{Ph} \\ \text{Ph} & & & \\ & & & \\ \text{Ph} & & & \\ & & & \\ \text{H}_3\text{C} & & & \\ & & & \\ \end{array}$$

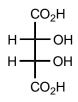
C)

D)

$$Ph$$
 $H_3C$ 
 $Ph$ 

E) no reaction

- 6. Which of the following Fischer projections represents (2R,3R)-tartaric acid? Hint: The OH is priority 1 and the CO<sub>2</sub>H is priority 2 for both chiral centers.
  - A)



B)

$$CO_2H$$
 $H \longrightarrow OH$ 
 $HO \longrightarrow H$ 
 $CO_2H$ 

C)

$$\begin{array}{c} \text{CO}_2\text{H} \\ \text{HO} \longrightarrow \text{H} \\ \text{HO} \longrightarrow \text{H} \\ \text{CO}_2\text{H} \end{array}$$

D)

$$\begin{array}{c} \text{CO}_2\text{H} \\ \text{HO} \stackrel{}{\longleftarrow} \text{H} \\ \text{H} \stackrel{}{\longleftarrow} \text{OH} \\ \text{CO}_2\text{H} \end{array}$$

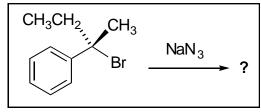
E) none of the above

7. What would be the major product of the following  $E_2$  reaction?

$$\begin{array}{c} \text{Br} \\ \text{CH}_3 - \text{CH}_2 - \text{CH} - \text{CH}_3 \end{array} \xrightarrow{\text{KOH}} ?$$

A) H<sub>3</sub>C

- B) OH
  - CH<sub>3</sub>-CH<sub>2</sub>-CH-CH<sub>3</sub>
- C)  $H_{3C} = C H_{3}$   $CH_{3}$
- D)  $CH_3-CH_2-CH=CH_2$
- E) O II CHo-CHo-C-CHo
- 8. The **major** product of the following reaction conditions will result from (Hint: chiral starting compound is  $3^{\circ}$  alkyl halide and the Na<sup>+ -</sup>N<sub>3</sub> is S<sub>N</sub>2 conditions):



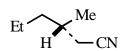
- A)  $S_N 2$
- B)  $S_N 1$
- C) E2
- D) E1
- E) there is no way to know

9. Predict the major product of the following  $S_{\rm N}2$  reaction:

A)



B)



C)

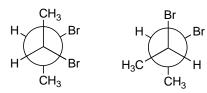


D)

E)

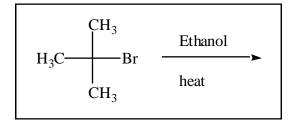
- 10. Which of the following can be used to synthesize (R)-2-cyanopentane from (R)-2-bromopentane?
  - A) NaBr
  - B) NaCN
  - C) NaI followed by KCN
  - D) NaCN followed by HI

### 11. How are the following compounds related?

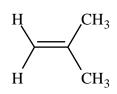


- A) Diastereomers
- B) Enantiomers
- C) Meso compounds
- D) Not related

#### 12. Circle the product of the following $E_1$ reaction:



A)



B)

$$H_3C$$
  $\longrightarrow$   $CH_3$   $CH_3$   $CH_3$ 

C)

$$H_3C$$
  $\longrightarrow$   $OE$ 

D)

$$H_3C$$
— $CH$ = $CH$ — $CH_3$ 

E)

$$H_3C$$
  $H_3$   $H_3C$   $H_2OEt$ 

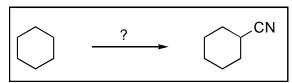
13. Which of the bases below would be best to accomplish the following reaction? Hint: Think E<sub>2</sub> and size of base!

- A) CH<sub>3</sub>O<sup>-</sup>Na<sup>+</sup>
- B) CH<sub>3</sub>CH<sub>2</sub>O<sup>-</sup>Na<sup>+</sup>
- C) (CH<sub>3</sub>)<sub>2</sub>CHO<sup>-</sup> Na<sup>+</sup>
- D) (CH<sub>3</sub>)<sub>3</sub>CO<sup>-</sup>Na<sup>+</sup>
- E) Na<sup>+</sup>-OH
- 14. Label the following carbons as either (R) or (S). <sup>2</sup>H is actually Deuterium and higher in priority than <sup>1</sup>H but lower than a methyl group.

- A) A = R, B = R, C = R, D = R, E = R
- B) A = S, B = S, C = S, D = S, E = S
- C) A = S, B = R, C = S, D = S, E = S
- D) A = S, B = S, C = R, D = S, E = S
- E) A = S, B = S, C = S, D = R, E = S
- 15. What would be the proper name of the following?

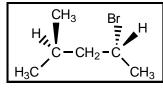
- A) (1R,2R)-trans-1,2-cyclohexanediol
- B) (1*R*,2*S*)-trans-1,2-cyclohexanediol
- C) (1S,2R)-trans-1,2-cyclohexanediol
- D) (1S,2S)-trans-1,2-cyclohexanediol
- E) (1S,2R)-cis-1,2-cyclohexanediol

16. What reactants are required to achieve the following transformation?



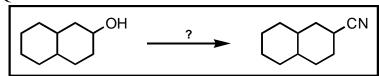
- A)
- 1. PBr<sub>3</sub> ➤
- 2. NaCN
- B)
- 1. NaCN
- $2.~\mathrm{H}_2\mathrm{SO}_4$
- C)
- 1. Br<sub>2</sub>, hv
  2. HCN
- D)
- 1. SOCl<sub>2</sub>
- 2. KCN
- E)
- 1.  $Br_2$ , hv
- 2. KCN

17. What would be the complete name of the following?



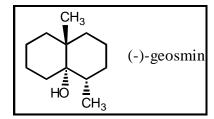
- A) (2R,4S)-2-bromopentane
- B) (R)-2-bromo-4-methylpentane
- C) (S)-4-bromo-2-methylpentane
- D) (2R,4R)-2-bromo-4-methylpentane
- E) (*S*)-2-bromo-4-methylpentane

18. What reactants are required to achieve the following transformation? Hint: Go to Question 9 for assistance.

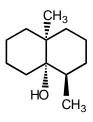


- A)
- NaCN
- B)
- 1) NaOH 2) KCN
- C)
- 2. KCN D)
- 1)  $Br_2/light$ 
  - 2) NaCN

19. The structure of (-)-geosmin is shown below. Which structure would be that of its enantiomer, (+)-geosmin?



A)



B)

$$\bigcap_{\mathsf{HO}}^{\mathsf{CH}_3}$$

C)

D)

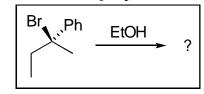
E) none of the above

20. Which of the following molecules have the *S* configuration? Hint: In II, the F has priority 1 ranking, the tert-butyl 2 ranking and the olefin 3 ranking.

- A) I, II
- B) I, III
- C) III, IV
- D) I, II, IV
- E) all of the above

### **Bonus Questions (2 X)** Points per Question = 5 PTS

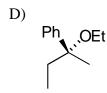
21. Predict the **major** product of the following  $S_N1$  solvolysis reaction:



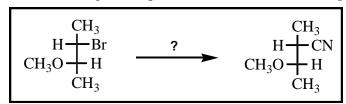
A) EtO Ph







22. Indicate the reagents required to achieve the following transformation:



A)



- B)
- 1) NaF 2) KCN
- C)
- 1) Nal
- 2) NaCN
- D)
- 1) NaOMe
- 2) HCN / light

## **Answer Key**

- 1. D
- 2. D
- 3. C
- 4. C
- 5. D
- 6. B
- 7. A
- 8. C
- 9. C
- 10. C
- 11. C
- 12. A
- 13. D
- 14. D
- 15. A
- 16. C/E
- 17. B
- 18. C
- 19. C
- 20. A
- 21. E
- 22. C